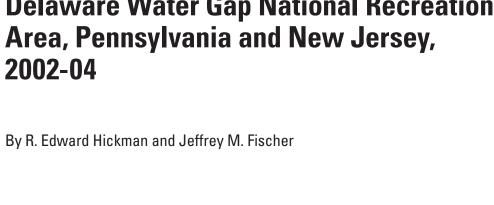


Prepared in cooperation with the National Park Service

Water Quality of Streams in and near the Delaware Water Gap National Recreation Area, Pennsylvania and New Jersey, 2002-04

Scientific Investigations Report 2007-5290

Water Quality of Streams in and near the **Delaware Water Gap National Recreation**



Prepared in cooperation with the National Park Service

Scientific Investigations Report 2007-5290

U.S. Department of the Interior DIRK KEMPTHORNE, Secretary

U.S. Geological Survey

Mark D. Myers, Director

U.S. Geological Survey, Reston, Virginia: 2008

For product and ordering information: World Wide Web: http://www.usgs.gov/pubprod Telephone: 1-888-ASK-USGS

For more information on the USGS—the Federal source for science about the Earth, its natural and living resources, natural hazards, and the environment:

World Wide Web: http://www.usgs.gov

Telephone: 1-888-ASK-USGS

Suggested citation:

Hickman R.E., and Fischer J.M., 2008, Water quality of streams in and near the Delaware Water Gap National Recreation Area, Pennsylvania and New Jersey, 2002-04: U.S. Geological Survey Scientific Investigations Report 2007-5290, 65 p.

Any use of trade, product, or firm names is for descriptive purposes only and does not imply endorsement by the U.S. Government. Use of company names is for identification purposes only and does not imply responsibility.

Although this report is in the public domain, permission must be secured from the individual copyright owners to reproduce any copyrighted material contained within this report.

Contents

Abstract		1
Introduction		2
Purpose	and Scope	2
Descript	ion of the Study Area	4
Previous	Investigations	4
Design and M	ethods of Study	4
Collectio	n of Water-Quality and Land-Use Data	4
Field and	Laboratory Methods	6
Analyses	s of Data	11
Mo	dification of Values Reported by Laboratory	11
Cal	culation of Medians with Nondetect Values	11
Cre	ation of Boxplots with Nondetect Values	11
Sun	nmary of Water-Quality Values in Blanks and Replicates	11
Rela	ations Between Water-Quality and Basin Characteristics	12
Equ	ations Relating Water-Quality Characteristics, Streamflow, and Season	12
Var	ation of Water Quality with Streamflow	12
Var	ation of Water Quality with Season	13
Det	ection of Future Changes in Water Quality	13
Water Quality	of Streams in and near the Delaware Water Gap National Recreation Area,	
2002-0	4	15
Summar	y of Water-Quality Characteristics at Each Station	15
Results o	of Analyses of Blanks and Replicates	15
Variation	of Water Quality with Basin Characteristics	18
Linear Ed	quations Relating Water Quality, Streamflow, and Season	23
Variation	of Water Quality with Streamflow and Season	23
	n of Future Changes in Water Quality	
Summary and	Conclusions	31
References C	ited	34
Appendixes—	-	
1.	Methods used to determine flow at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., at times of water-quality-sample collection, 2002-04	36
2.	A brief description of Tobit regression	
3.	Summary statistics for water-quality characteristics in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04	
4.	Equations relating water quality, streamflow, and season at stations on stream in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04	

Figures

1-2.	Map	os showing—
	1.	The Delaware Water Gap National Recreation Area and vicinity, Pa. and N.J
	2.	Stream water-quality monitoring stations and associated drainage basins in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-045
3-14.	Gra	phs showing—
	3.	Land use in drainage basins associated with water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-048
	4.	Distribution of values of (A) pH, acid-neutralizing capacity, specific conductance, dissolved calcium, (B) attenuation turbidity, total phosphorus, dissolved oxygen, total nitrogen, dissolved nitrate plus nitrite, and dissolved chloride at each water-quality monitoring station on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
	5.	Median concentrations of (A) total phosphorus and (B) total nitrogen at water-quality monitoring stations as a function of agricultural land use and the presence of permitted wastewater facilities in associated drainage basins, for streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
	6.	Median concentrations of total phosphorus and total nitrogen at water-quality stations as a function of wetland area and the presence of permitted wastewater facilities in associated drainage basins, for streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-0422
	7.	Median values of attenuation turbidity at water-quality monitoring stations as a function of wetland area in the associated drainage basins, for streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
	8.	Concentration of dissolved nitrate plus nitrite as a function of (A) streamflow and (B) day of the year at the water-quality monitoring station (01438754) on Adams Creek below Long Meadow Brook near Edgemere, Pa., near the border of the Delaware
		Water Gap National Recreation Area, Pa. and N.J., 2002-0425
	9.	Total phosphorus as a function of (A) streamflow and (B) day of the year at the water-quality monitoring station (01438892) on Dingmans Creek above Dingmans Falls near Dingmans Ferry, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
	10.	Total nitrogen concentration as a function of (A) streamflow and (B) day of the year at the water-quality monitoring station (01439830) on Big Flat Brook at Tuttles Corner, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
	11.	Attenuation turbidity as a function of streamflow at the water-quality monitoring station (01438700) on the Raymondskill Creek near Milford, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04

12.	Dissolved chloride as a function of (A) streamflow and (B) day of the year at the water-quality monitoring station (01439570) on the Sand Hill Creek at Bushkill, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-0428
13.	Values of pH as a function of streamflow at the water-quality monitoring station (01439680) on the Little Bush Kill at Bushkill, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J.,2002-0429
14.	Dissolved oxygen concentration as a function of streamflow at the water-quality monitoring station (01438700) on the Raymondskill Creek near Milford, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., July-September, 2002-04

Tables

1.	Water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
2.	Land-use characteristics of basins associated with water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
3.	Number of measurements of physical characteristics, plant nutrients, major ions, and organic compounds at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-049
4.	Selected water-quality characteristics measured at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-0410
5.	Summary of results of analyses of blanks and replicates from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
6.	Results of correlation tests between median values of water-quality characteristics at each water-quality monitoring station and the land use in the drainage basin, for streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
7.	Identification of increases and decreases in values of selected water-quality characteristics with increasing streamflow, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-0424
8.	Identification of water-quality characteristics with values that varied with season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04
9.	Minimum detectable differences between current (2002-04) and future water-quality values for stations on streams in and near the Delaware Water Gap National Recreation Area, Pa, and N.J., 2002-04, assuming 10 future measurements

Conversion Factors and Water-Quality Units

Inch/Pound to SI

Multiply	Ву	To obtain
	Area	
square mile	2.590	square kilometer
	Flow rate	
cubic feet per second (ft³/sec)	0.02832	cubic meter per second

Temperature is given in degrees Celsius (°C), which can be converted to degrees Fahrenheit (°F) by the following equation:

$$F = 1.8(C) + 32$$

Water-Quality Units

Concentrations of some constituents in water samples are given in milligrams per liter (mg/L), which is equivalent to "parts per million".

Water Quality of Streams in and near the Delaware Water Gap National Recreation Area, Pennsylvania and New Jersey, 2002-04

By R. Edward Hickman and Jeffrey M. Fischer

Abstract

Water samples were collected during 2002-04 at monitoring stations on 14 streams either within or entering the Delaware Water Gap National Recreation Area. The samples were collected from April through December of each year, mostly under low (base-flow) conditions, and were analyzed for major ions and nutrients (nitrogen and phosphorus). Results of the analyses, in concert with land-use information in the drainage basins associated with the samples, were used to define water-quality characteristics; to identify relations among water quality, streamflow, and season; and to establish a baseline and develop a method that could be used to detect future changes in water quality.

For a given water-quality characteristic, median values commonly varied among the 14 water-quality monitoring stations. For example, the median concentration of total phosphorus at the station on Sand Hill Creek (0.033 milligrams per liter as P) was four times the corresponding median concentration at the station on Vancampens Brook (0.008 milligrams per liter as P).

Results of correlations between median values of waterquality characteristics and land-use characteristics of the drainage basins indicate that agricultural practices and the presence of wetlands could be important factors affecting the concentrations of total nitrogen and total phosphorus in these streams. Results of analyses of samples from the nine stations without permitted wastewater facilities in their basins indicate that medians of both total phosphorus and total nitrogen increased with an increase in the area of agricultural land in the drainage basins; the levels of significance are 0.01 for total phosphorus and 0.01 for total nitrogen. When only the seven stations without permitted wastewater facilities and with less than 5 percent of the basin in agricultural land are considered, median concentrations of total phosphorus and total nitrogen increased with an increase in the area of wetlands in the basins; the levels of significance are 0.003 for total phosphorus and 0.03 for total nitrogen.

Linear equations between values of each water-quality characteristic at a station, streamflow, and season were developed by use of Tobit regression. The variations of water quality with streamflow and with season were identified from these equations.

Concentrations of total phosphorus, total nitrogen, and attenuation turbidity increased with increasing streamflow at more stations than concentrations decreased with increasing streamflow. Concentrations of dissolved orthophosphate phosphorus, dissolved nitrate plus nitrite, dissolved ammonia, and major ions decreased with increasing streamflow at more water-quality stations than concentrations increased with increasing streamflow.

Most water-quality characteristics varied with season at most stations due to reasons other than the seasonal variation in streamflow. Concentrations of total phosphorus and total nitrogen during the summer (July-September) often exceeded concentrations during the spring (April-June) and fall (October-December). As one example, concentrations of total nitrogen at the monitoring station on Big Flat Brook are between 0.1 and 0.2 milligrams per liter as N in the spring and fall, but increase to between 0.2 and 0.3 milligrams per liter as N during the summer.

A method based on the linear equations relating water quality to streamflow and season was developed to detect differences in water quality between current (2002-04) and future conditions. Changes in water quality would be identified by detecting differences between the intercept of the equation with current water quality and the intercept of the corresponding equation with future water quality. The intercept represents an estimate of the water quality at a station with a streamflow of 1 cubic foot per second during a season in which the seasonal variation of water quality is minimal.

The method to detect future changes in water quality allows for an estimate of the minimum amount of change from current water quality (2002-04) that can be detected. For example, if 10 measurements are made in the future, the minimum detectable changes in total phosphorus or total nitrogen at any of the stations are 4-12 percent of the intercepts in equations with current water quality.

Introduction

The Delaware Water Gap National Recreation Area (DWGNRA) straddles the boundary between the States of Pennsylvania and New Jersey (fig. 1). The natural beauty of this region, combined with its proximity to two major metropolitan areas (New York and Philadelphia), draws more than 5 million visitors to the DWGNRA each year. Most of these visitors arrive in the summer for water-based recreation such as canoeing, kayaking, swimming, and fishing.

The counties adjacent to the Delaware Water Gap National Recreational Area have been undergoing considerable population growth. For example, during 2000-05, the estimated increases in population of Sussex County, New Jersey, and Pike County, Pennsylvania, were 6 percent and 22 percent, respectively (U.S. Census Bureau, 2006a). The population growth in Pike County is expected to continue; a 23-percent increase during 2005-10 is forecast by the Pike County Office of Community Planning (Michael Mrozinski, Pike County Office of Community Planning, written communication, 2006).

Future increases in population in the adjacent counties are likely to cause undesirable changes in water quality in the streams flowing into and through the DWGNRA. Concentrations of nitrogen, phosphorus, and dissolved chloride in the streams may increase with increasing urban development due to increased discharge from wastewater treatment plants and (or) septic systems. These changes in water quality may, in turn, produce undesirable changes in the aquatic biota in these streams; algae may increase and populations of pollution-intolerant fish may be reduced.

The Delaware River Basin Commission (DRBC), which oversees the management of the water resources within the Delaware River Basin, has recognized the need to protect the exceptional water quality and ecology of the tributaries of the Delaware River within DWGNRA. In 1992, the DRBC designated these waters "Outstanding Basin Waters" and enacted the "Special Protection Water Regulations." These regulations prohibit human development from causing any measurable change in water quality in the streams draining to the Delaware River within the DWGNRA.

To use these anti-degradation regulations to protect the water quality of the DWGNRA tributaries, three types of information are needed. First, sufficient data are needed to describe current water quality. Although water-quality data have been collected by Pike and Monroe counties in Pennsylvania, the States of Pennsylvania and New Jersey, the DRBC, and the NPS, these data are not considered sufficient to define current water quality. Many streams were not sampled or sampled only quarterly. Methods differed from agency to agency. Some laboratory methods had high reporting levels resulting in many nondetect values.

Second, there needs to be a method to indicate whether the future water quality will be different from the current water quality. Water quality often varies with streamflow and season. For instance, concentrations of phosphorus and nitrogen in streams typically are elevated during high spring and summer streamflow following fertilizer applications; alternately, concentrations in streams downstream from metropolitan areas may be highest during periods of low flow, when contributions from point sources are greater relative to streamflow, and dilution is less (U.S. Geological Survey, 1999). Therefore, a method is more likely to detect changes between current and future water quality if the variations of water quality with stream and season are accounted for.

Third, there needs to be a plan to monitor water quality in the future to allow detection of changes between current and future water quality. This plan should be designed on the basis of the method used to detect changes between current and future water quality.

To address these issues, the DWGNRA received funding from the NPS-USGS Water Quality Assessment and Monitoring Partnership for a project to provide baseline water-quality data and an understanding of water-quality in streams entering the Delaware Water Gap National Recreation Area. Objectives of this investigation were

- Collect samples of water in 14 streams within or entering the Delaware Water Gap National Recreation Area during selected seasons in 2002-03 and analyze them for concentrations of plant nutrients and major ions; the period of data collection was changed to 2002-04 because of drought conditions in 2002.
- Provide an overall assessment of the current and potential effects of development on long-term and episodic water quality in streams.
- Develop a cost effective strategy for future monitoring.

Purpose and Scope

The purpose of this report is twofold. First, selected results of the analyses of water samples collected at monitoring stations on 14 streams in and near the Delaware Water Gap National Recreation Area during the spring, summer, and fall of 2002-04 are presented. The water-quality properties and constituents discussed in this report are dissolved oxygen, specific conductance, pH, attenuation turbidity, phosphorus and nitrogen species, acid-neutralizing capacity, dissolved calcium, and dissolved chloride.

Some basin characteristics that appear to affect water quality, such as land use, geology, and the presence of permitted wastewater facilities, are identified. Also identified was whether values of a water-quality characteristic at a station (a) increased or decreased with increasing streamflow, and (or) (b) changed with the season. Summary statistics and boxplots are provided for each water-quality characteristic at each monitoring station.

Second, a method to detect differences between current water quality and water quality measured in the future

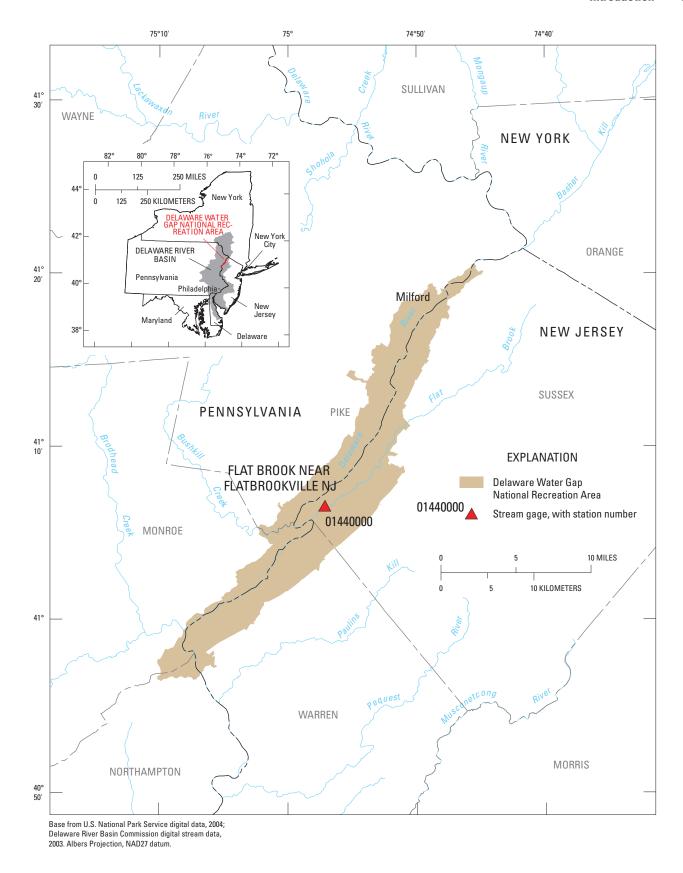


Figure 1. The Delaware Water Gap National Recreation Area and vicinity, Pa. and N.J.

is described. Estimates of the minimum amount of change in each water-quality characteristic that can be detected are presented.

Description of the Study Area

The Delaware Water Gap National Recreation Area is a 69,300-acre park maintained by the National Park Service (NPS) along a 40-mile reach of the Delaware River (fig. 1). The park is mostly undeveloped; 81 percent is forested (fig. 2).

Numerous tributaries of the Delaware River pass through the park; some of these tributaries are shown in figure 2. Most of the drainage basins of these tributaries are outside of the park, and only the most downstream parts of the basins are within the park.

Most of the tributaries to the Delaware River within the DWGNRA have very high quality water, according to the National Park Service (National Park Service, 2006). Some of these tributaries support self-sustaining populations of eastern brook trout (*Salvelinus fontinalis*), a fish that survives only in the coldest and cleanest water (Eastern Brook Trout Joint Venture, 2006).

Previous Investigations

Available information suggests that concentrations of phosphorus and nitrogen are likely to increase due to an increase in discharges from wastewater treatment plants and (or) septic systems, as well as from other sources such as increased soil erosion and fertilizer use. The USGS (1999) summarized concentrations of phosphorus and nitrogen in selected streams throughout the Nation; in general, concentrations were greater in streams draining predominantly urban basins than in streams draining undeveloped basins. Fischer and others (2004) examined base-flow concentrations of total nitrogen and total phosphorus measured during May and June 1999-2001 in streams with drainage basins composed of 10 percent or less agricultural land use and located in the Delaware River Basin; they concluded that concentrations in streams with drainage basins composed of more than 10 percent urban land were greater than corresponding concentrations in streams with basins composed of less urban land They also noted that the percentage of all samples (base-flow and stormflow) in which phosphorus concentration exceeded 0.1 milligrams per liter as phosphorus increased with the percentage of basin composed of urban land. Heisig (2000) reported the results of base-flow samples collected on streams draining mostly unsewered basins in the Croton Watershed in New York State during 1996-97; Heisig concluded that concentrations of nitrate in streams under base-flow conditions increased with the density of unsewered housing in the basin.

Concentrations of dissolved chloride also may increase with increasing urban development due to increased discharge from wastewater treatment plants and (or) septic systems—chloride is commonly present in domestic wastewater—as

well as that washing off road surfaces. Fischer and others (2004) reported that concentrations of dissolved chloride measured in streams under base-flow conditions in the Delaware River Basin during May and June 1999-2001 increased with the road density in the basin. Heisig (2000) reported that chloride concentrations measured under base-flow conditions during 1996-97 in the Croton Watershed increased with the amount of salt applied to roads in the basins.

The effects of increasing urbanization, including increases in concentrations of plant nutrients and dissolved chloride in the streams, may cause undesirable changes in aquatic biota in these streams. From stream samples collected in the Delaware River Basin during 1998-2001, Fischer and others (2004) reported that EPT richness (the number of mayflies, stoneflies, and caddisflies) and habitat quality of the streams declined with increasing road density. The decline in EPT richness indicates that the relative abundance of pollution-sensitive benthic invertebrates decreased with increasing road density.

Design and Methods of Study

Current stream water quality conditions in the Delaware Water Gap National Recreation Area were determined on the basis of the collection and analysis of samples from 14 monitoring stations on tributaries to the Delaware River that pass through the Delaware Water Gap National Recreation Area. The data resulting from field measurements and from laboratory chemical analysis of water samples were summarized; analyzed by graphical and statistical methods to identify relations between water quality, streamflow, season, and (or) basin characteristics; and used to determine the minimum amounts of future change in properties and constituent concentrations at a station which could be detected by a described method.

Collection of Water-Quality and Land-Use Data

The 14 stations where samples were collected are described in table 1 and the locations are shown in figure 2. Stations were selected (1) to be near to or within the park, (2) to represent a range of development conditions, (3) to be suitable for water-quality sampling and streamflow measurement, and (4) to include as many streams as feasible.

Of these stations, 13 are located on the tributaries near where they enter the park (fig. 2, table 1). The basins associated with these stations range in size from 3.7 to 117 mi² and lie either entirely or almost entirely outside of the park. The 7.4-mi² drainage basin of the station on Vancampens Brook (station 01440100) lies entirely within the park.

The land-use data used to characterize these basins were obtained from the interpretation of satellite imagery collected during the early to mid 1990s (1992 National Land Cover Data), modified to represent the amount of urban land in 2000. A description of the 1992 National Land Cover Data

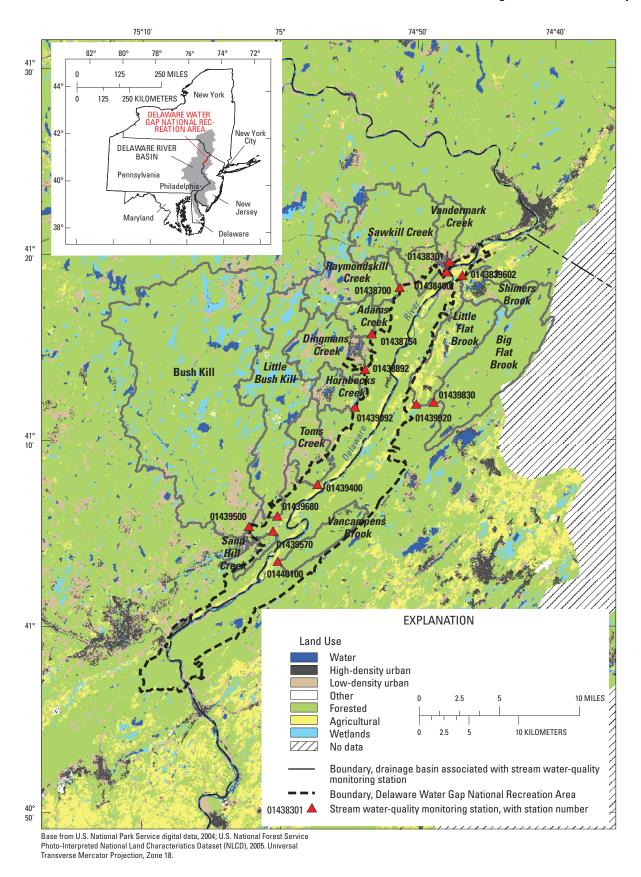


Figure 2. Stream water-quality monitoring stations and associated drainage basins in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

(NLCD92) can be found in a report by the U.S. Geological Survey (2006). Data in the NLCD92 were modified by Rachel Riemann (U.S. Forest Service, written commun., 2005) to effect two changes. On the basis of local road density levels, the classification of some pixels was changed from "undeveloped" to "urban." This had two effects—first, correcting for development not recognized because of leaf cover and procedures originally used to classify the imagery, and second, changing the extent of urban land depicted in the NLCD92 to that which existed in 2000. This was accomplished by merging NLCD92 information for the basins with local road density calculated from the 2000 TIGER Line dataset (U.S. Census Bureau, 2006b). Those pixels classified as "forested", "agriculture", or "transition" in the NLCD92 were reclassified as "low-intensity urban use" if the road density of the pixel was above a certain threshold. These techniques are described in Lister and others (2005).

The drainage basins associated with the stations are predominantly forested; forest land use makes up 63 percent or more of the area of each basin (table 2; fig. 3). High-density urban land accounts for a maximum of 8 percent of each basin; low-density urban land, a maximum of 22 percent; and agricultural land, a maximum of 18 percent.

As shown in table 2, some type of permitted wastewater facility is discharging to the basin of 5 of the 14 streams. This information was provided by the New Jersey Department of Environmental Protection (Thomas Cosmas, written commun., 2007) and obtained from the Pennsylvania Department of Environmental Protection website, accessed April 3, 2007, at

http://www.emappa.dep.state.pa.us/emappa/viewer.htm. The permitted discharges include discharges from septic systems.

Field and Laboratory Methods

Physical characteristics and concentrations of plant nutrients and major ions were either measured in the field by NPS and USGS personnel or determined from laboratory analysis of samples they collected (table 3). Values of all water-quality characteristics were stored in the USGS National Water Information System (NWIS) computer system (described in Mathey, 1998) and can be retrieved from the NWIS website http://waterdata.usgs.gov/usa/nwis/qw; these data have also been published in Deluca and others (2005, 2006). An additional sample collected at each station was analyzed for manmade organic compounds; results are presented in Deluca and others (2005) and are available from NWIS.

Field measurements were made and samples were collected for analyses during May-November of 2002 and April-December of 2003 and 2004 (table 3); there were no measurements or samples during January-March. Samples for plant nutrients were collected 12 to 15 times per year at each station; samples for major ions were collected at least 5 times per year (about every 6 weeks). Samples were collected over a range of flow conditions, but most were collected under "base-flow" conditions (times other than during or immediately following storms). During spring (March-May) and fall (October-December), samples were collected approximately

Water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., Table 1. 2002-04.

[Short name, shortened version of station name; latitude and longitude are in degrees, minutes, seconds; drainage area is in square miles; Cr, creek; Br, brook]

Short name	Station number	Station name	Latitude	Longitude	Drainage area
Adams Creek	01438754	Adams Creek below Long Meadow Brook near Edgemere, Pa.	411540	0745324	3.7
Big Flat Brook	01439830	Big Flat Brook at Tuttles Corner, N.J.	411200	0744855	28.3
Bush Kill	01439500	Bush Kill at Shoemakers, Pa.	410517	0750217	117
Dingmans Creek	01438892	Dingmans Creek above Dingmans Falls near Dingmans Ferry,Pa.	411347	0745350	13.9
Hornbecks Creek	01439092	Hornbecks Creek at Emery Road near Dingmans Ferry, Pa.	411145	0745436	6.4
Little Bush Kill	01439680	Little Bush Kill at Bushkill, Pa.	410552	0750015	32.6
Little Flat Br	01439920	Little Flat Brook at Peters Valley, N.J.	411154	0745009	14.7
Raymondskill Cr	01438700	Raymondskill Creek near Milford, PA.	411811	0745121	20.4
Sand Hill Creek	01439570	Sand Hill Creek at Bushkill, Pa.	410506	0750032	3.5
Sawkill Creek	0143839602	Sawkill Creek 1000 ft above Mouth at Milford, Pa.	411900	0744759	24.7
Shimers Brook	01438400	Shimers Brook near Montague, N.J.	411847	0744651	7.1
Toms Creek	01439400	Toms Creek at Egypt Mills, Pa.	410733	0745720	9.3
Vancampens Br	01440100	Vancampens Brook near Millbrook, N.J.	410328	0750012	7.4
Vandermark Cr	01438301	Vandermark Creek at 4th Street at Milford, Pa.	411930	0744749	5.2

Table 2. Land-use characteristics of basins associated with water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Data are from the USGS National Land Cover data for 1992 (U.S. Geological Survey, 2006), after modification by Rachel Riemann (U.S. Forest Service; see discussion in text) to identify low-density urban land use; information indicating the presence of a permitted wastewater facility discharging to basins was obtained from a communication from the New Jersey Department of Environmental Protection (Thomas Cosmas, written commun., 2007) and from information on the Pennsylvania Department of Environmental Protection website accessed April 3, 2007, at http://www.emappa.dep.state.pa.us/emappa/viewer.htm]

			Land use, in		rainage basir CD92 land use		0.1 percent		There is a per-
Short name	Station number	High- density urban (21, 22, 23, 85)	Low- density urban (27, 28, 29)	Agricul- tural (81, 82)	Forested (41, 42, 43)	Wet- lands (91, 92)	Water (11)	Other (31, 32, 33)	mitted waste- water facility discharging to basin
Adams Creek	01438754	2.3	8.9	1.1	81.5	3.4	1.8	1	No
Big Flat Brook	01439830	0.4	0.9	0.4	93.2	2.2	1.2	1.7	No
Bush Kill	01439500	0.5	5.4	0.4	81.9	9.4	2.2	0.2	Yes
Dingmans Creek	01438892	4.2	19.4	0.2	64.	6.1	6	0.1	No
Hornbecks Creek	01439092	5.5	19.3	2.8	62.5	5.5	4.4	0	No
Little Bush Kill	01439680	0.9	6.5	0.6	73.5	14.7	3.2	0.6	Yes
Little Flat Br	01439920	3.1	2.4	18.4	69.6	5.7	0.7	0	No
Raymondskill Cr	01438700	1.3	13.2	1.1	69.9	11.2	3.2	0.1	No
Sand Hill Creek	01439570	6	11.8	3.2	69.4	6.1	3.5	0	Yes
Sawkill Creek	0143839602	2	8.1	1.1	83.5	4.3	1	0	Yes
Shimers Brook	01438400	8.5	6.9	5.3	70.4	4.7	4.3	0	No
Toms Creek	01439400	1.2	21.5	0.5	74.4	1.5	0.7	0.2	No
Vancampens Br	01440100	3.2	1	0.3	92.8	1.7	1.1	0	No
Vandermark Cr	01438301	3.7	7.6	0.6	87.3	0.6	0.2	0	Yes

every 4 weeks. During summer (June-September), samples were collected every 2 weeks. At least one additional sample was collected under "stormflow" conditions (during or immediately following rainstorms) each year at each site.

Measurements of plant nutrients and major ions (table 4) were made on composite water samples, which were collected in a 3-liter polyethylene bottle at multiple points across the stream; collection techniques are described by Shelton (1994). The majority of samples for laboratory analysis were collected by personnel wading the stream and using either an open bottle or a DH-81 sampler (Federal Interagency Sedimentation Project, undated); a few samples were collected from a bridge by lowering an open bottle on a line. All equipment was cleaned prior to the collection of each sample.

Subsamples to be sent to the laboratory for determination of concentrations of nutrients and major ions were obtained from the 3-liter bottle after it was shaken. Subsamples for dissolved constituents were filtered through a 0.45-micrometer

membrane filter. All samples were chilled prior to transport to the laboratory.

Two types of quality-assurance/quality-control (QA/QC) samples for plant nutrients and major ions were collected—blanks and replicates. A blank was created by putting deionized water through the equipment used for sample collection, compositing, and filtration. A replicate of a selected environmental sample was obtained by collecting a second environmental sample using the same sampling protocols. Blanks and replicates were processed by the same methods used for environmental samples and then chilled and sent to the laboratory for analyses.

Water temperature, dissolved oxygen, pH, and specific conductance were measured directly in each stream by use of field meters (table 4). The values for water from the center of the stream are reported. All values of dissolved oxygen concentration in this report are expressed as percent of saturation; values are expressed in this way to investigate effects other than temperature on dissolved oxygen.

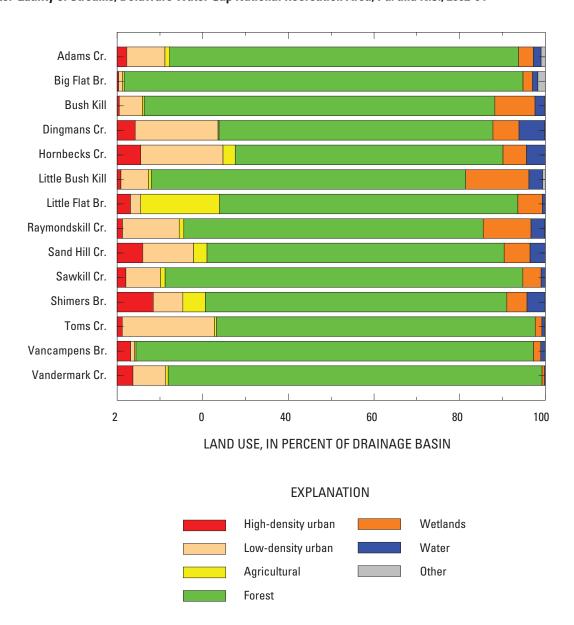


Figure 3. Land use in drainage basins associated with water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04. [Cr., Creek; Br., Brook]

Attenuation turbidity was measured in a sample collected at the center of each stream. Attenuation turbidity values in this report were determined with a LaMotte Colorimeter, which determines turbidity as function of the amount of incident light absorbed by the water sample (LaMotte Company, 1997); the light detector is 180 degrees to the incident light source. This is a nonstandard method to measure turbidity; the standard method is to determine turbidity from measurements of the amount of light scattered at 90 degrees to the incident light source (American Public Health Association, 1998). The values measured by the LaMotte Colorimeter are affected by both the color of the water and the concentration of particles (Tim Parent, LaMotte Company, written commun., 2007). The term "attenuation turbidity" is used throughout this report

to remind the reader that these values were collected with a nonstandard method.

Streamflows at the times of sample collection were measured or estimated at 13 of the 14 water-quality monitoring stations; methods are described in appendix 1. Streamflows associated with samples collected from Vandermark Creek at 4th Street at Milford, PA, (station 01438301) were not determined or estimated because this station was added at the last moment prior to sampling and because a site suitable for streamflow measurement could not immediately be found.

Laboratory methods used to analyze the water samples are listed in table 4. The concentration of each plant nutrient or major ion reported here was determined at the USGS National Water-Quality Laboratory in Denver, Colo. Three types of

Table 3. Number of measurements of physical characteristics, plant nutrients, major ions, and organic compounds at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Measurements were either made in the field or from laboratory analysis of samples collected in the field]

Water-quality characteristic	Average number of measurements per station
Physical characteristics (including streamflow)	
Streamflow	(See Appendix 1)
Specific conductance	43
рН	43
Water temperature	44
Dissolved oxygen, in milligrams per liter	43
Dissolved oxygen, in percent of saturation	42
Barometric pressure	44
Attenuation turbidity	42
Suspended sediment	13
Plant nutrients	
Total phosphorus	44
Dissolved orthophosphate phosphorus	43
Total nitrogen	19
Total organic nitrogen plus ammonia	25
Dissolved ammonia	43
Dissolved nitrate plus nitrite	43
Dissolved nitrite	43
Major ions	
Acid-neutralizing capacity	18
Dissolved aluminum	6
Dissolved boron	18
Dissolved calcium	18
Dissolved chloride	18
Dissolved fluoride	11
Dissolved magnesium	18
Dissolved silica	18
Dissolved sodium	18
Dissolved sulfate	18
Residue of dissolved constituents upon evaporation	18
Organic compounds commonly found in wastewater	
The 67 compounds, including surfactants and their degradates, food additives, fragrances, antioxidants, flame retardants, plasticizers, industrial solvents, disinfectants, fecal sterols, polycyclicaromatic hydrocarbons, and high-use domestic pesticides are listed in Zaugg and others (2002)	1

concentration values were reported; each concentration value consists of a remark code and a numeric value. An uncensored value (remark code is a blank) indicates that the concentration was reliably determined to be greater than or equal to the laboratory reporting level. An estimated value (remark code = "E") indicates that the analyte was detected but could not be reliably quantified by the laboratory. A nondetect value (remark code = "<") indicates that the concentration is less than the reporting level. For a given constituent, the reporting

level sometimes changed during the period of study as a result of changes in methods, reporting protocol, or the presence of substances in the sample that interfered with the results of analysis.

Selected water-quality characteristics measured at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Physical characteristics were measured in the field; measurements of plant nutrients and major ions were determined from laboratory analysis of samples collected in the field; mg/L, milligrams per liter; P, phosphorus; N, nitrogen; CaCO₃; calcium carbonate; Cl, chloride; Ca, calcium; n.a., not applicable]

Characteristic	Laboratory method	Comment
	Physical characterist	ics
Instantaneous streamflow, in cubic feet per second	n.a.	Determined for 13 of the 14 stations using a variety of methods (see Appendix 1)
Dissolved-oxygen concentration, in percent saturation	n.a.	Calculated by use of equation provided by R.J. Pickering (U.S. Geological Survey, written commun., 1981) and from field measurements of water temperature, specific conductance, dissolved oxygen in mg/L, and barometric pressure; measurements of dissolved oxygen in mg/L were made with a YSI 54A
Specific conductance, in microsiemens per centimeter at 25 degrees Celsius	n.a.	Measured with a YSI 30
pH, in standard units	n.a.	Measured with an Oakton pH Testr 2
Attenuation turbidity, in attenuation units	n.a.	Measured with a LaMotte Smart Colorimeter; values reflect true color as well as particles
	Plant nutrients	
Total phosphorus, in mg/L as P	EPA 365.1 in Methods for the Determination of Inorganic Sub- stances in Environmental Samples (EPA/600/R-93/100)	None
Dissolved orthophosphate phosphorus, in mg/L as P	Fishman (1993)	None
Total nitrogen, in mg/L as N	Patton and Kryskalla (2003)	Measurements began 10/1/2003
Total organic nitrogen plus ammonia, in mg/L as N	Patton and Truitt (2000)	Measurements ended 9/30/2003
Dissolved nitrate plus nitrite, in mg/L as N	Fishman (1993)	None
Dissolved ammonia, in mg/L as N	Fishman (1993)	None
	Major ions	
Acid-neutralizing capacity, in mg/L as CaCO ₃	Fishman and Friedman (1989)	None
Dissolved calcium, in mg/L as Ca	Fishman (1993)	None
Dissolved chloride, in mg/L as Cl	Fishman and Friedman (1989)	None

Analyses of Data

Methods discussed in this section include those used to (1) calculate medians and create boxplots for datasets with nondetect values, and (2) determine if values of physical properties or chemical constituents increased or decreased with changes in streamflow, season, or selected basin characteristics. A method to identify future changes in water quality is presented, and the amount of change which can be detected is discussed.

Modification of Values Reported by Laboratory

For some of the nutrients and major ions discussed in this report, the USGS National Water Quality Laboratory set reporting levels of nondetect values to twice the method detection limit. This practice was used to minimize the probability of having a false negative and is discussed in Oblinger-Childress and others (1999). According to Helsel (2005), however, this doubling of the detection level produces bias when the values are interpreted. For that reason, the reporting levels for these nondetect values were set back to the method detection level (half the reporting level from the laboratory). This action was suggested by Helsel (2005) as one way to avoid bias in the interpretation of these values.

For samples collected prior to October 1, 2003, concentrations of total nitrogen were calculated by summing concentrations of total organic nitrogen plus ammonia and dissolved nitrate plus nitrite. A reporting level of 0.1 mg/L for the calculated values of total nitrogen was selected so as to be greater than the reporting levels for total organic nitrogen plus ammonia (0.05 mg/L as N) or dissolved nitrate plus nitrite (0.025 or 0.03 mg/L as N) after reporting levels were halved. Below is the set of rules under which values of total nitrogen were calculated.

- Estimated concentrations of total organic nitrogen plus ammonia and dissolved nitrate plus nitrite were considered uncensored values.
- (2) For the purposes of calculating total nitrogen, nondetect concentrations were set to 0.
- (3) Concentrations of total organic nitrogen plus ammonia and dissolved nitrate plus nitrate were summed.
 - a) If the sum was greater than or equal to 0.1 mg/L as N, the concentration of total nitrogen was considered to be uncensored and equal to the sum.
 - b) If the sum was less than 0.1 mg/L as N, the concentration of total nitrogen was set to "<0.1 mg/L as N."</p>

Calculation of Medians with Nondetect Values

If nondetect values accounted for less than 50 percent of the set of measurements for a water-quality characteristic at a station, the median of all values was determined without regard to the remark codes. All of these medians were determined to be greater than or equal to the greatest reporting level of the nondetect values.

If nondetect values accounted for 50 to 80 percent of a set of measurements, the median was estimated by use of the regression on order of statistics (ROS) method described in Helsel (2005). Calculations were done with a computer program obtained from the "Practical Statistics for the Sciences" website accessed July 19, 2005, at http://www.practicalstats.com/index.html.

Medians were not estimated for dissolved orthophosphate phosphorus and dissolved ammonia measured for some stations because more than 80 percent of the values were nondetect values. For these stations, the medians are shown as being less than the greatest reporting level of the nondetect values.

Creation of Boxplots with Nondetect Values

The following rules were used to create boxplots of the measurements of each water-quality characteristic at each station when the measured values included nondetect values.

- (1) If 5 percent or less of the set of measurements at all stations were nondetect values, then all values were treated as uncensored.
- (2) If more than 5 percent of the set of measurements at any station were nondetect values, then
 - the greatest reporting level of nondetect values at all stations was determined and shown in the figure, and
 - b) boxplots of values for all stations were censored at this greatest reporting level.

Summary of Water-Quality Values in Blanks and Replicates

A summary of concentrations of nutrients and major ions in the blanks and replicate/sample pairs (each pair is composed of a replicate and the associated sample) is presented. For blanks, the summary includes the number of blanks in which the compound was detected, the greatest concentration detected, and the greatest reporting level of nondetect concentrations. For replicate/sample pairs, the summary includes (a) the number of replicate/sample pairs with incongruous results and (b) the median and maximum differences between concentrations in replicates and corresponding samples. Incongruous results are considered to exist if a compound was detected in only the sample or in only the replicate, but not in both. Dif-

ferences between concentrations in replicates and corresponding samples were not calculated for those replicate/sample pairs with incongruous results; a difference between nondetect concentrations in both sample and replicate was recorded as zero.

Relations Between Water-Quality and Basin Characteristics

The Kendall's tau method (described in Helsel and Hirsch, 1992) was used to determine whether, for a selected water-quality characteristic, median values increased or decreased with increasing values of either (1) land use in the associated drainage basins, or (2) median values of other water-quality characteristics; calculations were done with the Corr procedure (SAS Institute, Inc., 1999a). This nonparametric method measures the strength of monotonic correlation between two variables; an increase or decrease in the selected water-quality characteristic was considered to exist if the level of significance of the correlation was 0.05. Correlations for each water-quality characteristic were calculated with only those stations for which uncensored median concentrations could be calculated.

Some relations identified by use of Kendall's tau were plotted. In each plot, a LOWESS curve shows a smoothed relation between the two variables. The method of calculation of LOWESS curves is briefly described in Helsel and Hirsch (1992).

Equations Relating Water-Quality Characteristics, Streamflow, and Season

Equations relating values of physical properties or chemical concentrations in a water sample from a station to streamflow and season took one of the two following forms:

$$C = B_0 + \left[B_1 * log(Q) \right] +$$

$$\left[(B_2 * sin(0.0172* jday)) + (B_3 * cos(0.0172* jday)) \right],$$
(1)

or

$$log(C) = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172* jday)) + (B_3 * cos(0.0172* jday)) \right],$$
 (2)

where

log = base-10 logarithm;

C = value of selected physical property or

chemical concentration;

 B_o = intercept;

 B_1 = coefficient for logarithm of streamflow;

Q = streamflow, in cubic feet per second;

 B_2 = first seasonal coefficient;

sin = sine;

jday = day of the year;

 B_3 = second seasonal coefficient; and

cos = cosine.

For each water-quality characteristic, the selection of whether to use equation 1 or equation 2 to relate water-quality values to streamflow and season was based on an examination of the plot of the data. Equation 1 was selected for dissolved oxygen concentration and pH; equation 2 was selected for all other water-quality characteristics.

The sine and cosine terms in equations 1 and 2 are the seasonal terms. The form of the seasonal terms was taken from Helsel and Hirsch (1992); two seasonal terms are needed to define the amplitude and the phase shift of the relation between water-quality characteristic and season.

Equations were developed by use of Tobit regression (briefly described in appendix 2) rather than least-squares regression, because in equations 1 and 2, Tobit regression allows the water-quality values to be nondetect values. If all values of a water-quality characteristic are detected, the results of Tobit regression are the same as for least-squares regression.

Equations were not developed if more than 80 percent of the measured values were "non-detects" and if there were fewer than 12 detected values. These requirements (that 20 percent of the water-quality values should be detected and that three or more detected values for each coefficient including the intercept were available) were suggested by David Lorenz (U.S. Geological Survey, written commun., 2007).

Equations were not reported if the assumptions of linearity, normality, and constant variance were not met. Linearity, constant variance, and normality of the data were verified by examining plots of data and residuals. How well the values predicted with each equation matched measured values was determined from an examination of plots of predicted and measured values as a function of streamflow and season. The effect of outlying points was considered. Plots of residuals as a function of streamflow and season were examined to determine whether there was constant variance. Probability plots of residuals were examined to determine if the residuals were normally or near normally distributed.

For 13 of the 14 stations, streamflow values were those instantaneous values measured or estimated at the time of sampling at the stations. The daily mean streamflows at Flat Brook near Flatbrookville, N.J., (01440000) on the days of sample collection were used in place of streamflows at the station on Vandermark Creek (01438301). No analysis was made to determine whether the daily streamflow in Flat Brook was representative of streamflow in Vandermark Creek.

Variation of Water Quality with Streamflow

Whether the value of a physical property or chemical constituent changed with increasing streamflow was deter-

mined from the value of B_1 in either equation 1 or equation 2. An increase was considered to exist if B_1 was greater than 0 at a 0.05 level of significance; a decrease was considered to exist if B_1 was less than 0 at a 0.05 level of significance.

Variation of Water Quality with Season

A nested F-test (described in Helsel and Hirsh, 1992) was used to determine whether the value of a physical property or chemical constituent varied with season for reasons other than the seasonal variation of streamflow. The outcome of the test was used to determine whether a "complex" equation with seasonal terms was significantly better than a "simple" equation without seasonal terms in explaining the variation in the values of a water-quality characteristic at a station. The "complex" equations are equations 1 and 2 previously discussed. The corresponding "simple" equations are

$$C = B_0 + B_1(\log(Q)), \tag{3}$$

and

$$log(C) = B_0 + B_1 (log(Q)), \tag{4}$$

where

log = base-10 logarithm;

C = value of selected water-quality

characteristic;

 B_0 = intercept;

 B_1 = coefficient for logarithm of

streamflow; and

Q = streamflow, in cubic feet per second.

One equation was used for all relations for each waterquality characteristic. Equation 3 was used for dissolved oxygen concentration and pH. Equation 4 was used all other water-quality characteristics.

The following equation for a nested F-test was taken from Helsel and Hirsch (1992).

$$F = \frac{(SSE_s - SSE_c) / (df_s - df_c)}{(SSE_c / df_c)},$$
(5)

where

F = test statistic,

 SSE_s = sum of squares of errors for simple

equation 3 or 4,

 SSE_c = sum of squares of errors for complex

equation 1 or 2,

 df_s = degrees of freedom of simple model, and

 df_c = degrees of freedom of complex model.

The degrees of freedom of the simple model and complex model are defined as

$$df_s = n - 2, (6)$$

and

$$df_{c} = n - 4 (7)$$

where

 $df_{s} =$ degrees of freedom of simple model,

 $d\hat{f}$ = degrees of freedom of complex model, and

n = number of measurements in model.

Equation 5, however, is formulated for use with least-squares regression and was modified for use with Tobit regression by substituting values of "scale" (see Appendix 2) in place of values of sum of squares of error by use of equation 8:

$$SSE \approx scale^2 \times df$$
, (8)

where

SSE = sum of squares of error,

scale = measure of "goodness of fit" from Tobit

regression, and

df = degrees of freedom.

Substituting equations 6 to 8 into equation 5 produces the equation given below. It is this equation that was used to determine the value of the test statistic F-test with Tobit regression.

$$F = \frac{((scale_s^2 \times (n-2)) - (scale_c^2 \times (n-4))) / (2)}{scale_c^2},$$
 (9)

where

F = test statistic,

scale = scale for simple equation 3 or 4,

scale = scale for complex equation 1 or 2, and

n = number of measurements.

For this report, the complex model was considered to be significantly better than the simple model in describing the variation of the water-quality values if the test statistic (F) equaled or exceeded the value of the F distribution with 2 and df_c degrees of freedom at a 0.05 level of significance. In these cases, the values of water quality were considered to vary with season.

Detection of Future Changes in Water Quality

Differences between current and future values of a waterquality characteristic are based on equations between values of that water-quality characteristic at a station, streamflow, and season. Separate equations are developed by use of Tobit regression for values of current water quality and water quality to be measured in the future. The form of the equation for

a selected water-quality characteristic at a station is shown either by equation 1 or equation 2.

Differences between current and future values of a waterquality characteristic at a station are based on a comparison of the intercepts in the two equations. The method of comparison of intercepts was provided by Gregory Schwarz (U.S. Geological Survey, written commun., 2005). The comparison is based on the following equation:

$$t = (B_{0p} - B_{0f}) / \left[\text{var}(delta(\text{intercepts})) \right]^{0.5}, \tag{10}$$

where

= test statistic,

= intercept of relation with current values of a water-quality characteristic at a

 B_{of} = intercept of relation with future value of a water-quality characteristic at a station, and

var(*delta*(intercepts)) = variance of the difference between the intercept with current water quality and the intercept with future water quality.

The two intercepts will be known, but the variance of the differences between intercepts will not. The variance term, however, can be approximated from the number of measurements of current and future water quality and the variance of the errors in the relation for current water quality using the equation

$$Var(delta(intercepts)) \approx ((m+n)/(m*n)) (Var(errors)),$$
 (11)

where

Var(delta(intercepts)) = variance of thedifference between intercepts, = number of current measurements, m = number of future measurements, and Var(errors) = variance of the errors in the relation with current water quality.

In turn, the variance of the errors of the relation can be approximated with the mean square error of the relation with current water quality using the equation

$$Var(errors) \approx MSE$$
, (12)

where

Var(errors) = variance of the errors in the relation,

MSE mean square error of relation with current water quality.

If values in equations 11 and 12 are substituted in equation 10 and scale is substituted for mean square error, equation 13 is the result. This equation can be used to determine whether the intercepts are significantly different from one

another. All values on the right hand side will be known after the future sampling.

$$t \approx (B_{0p} - B_{0f}) / \left[((m+n)/(m*n)) (scale^2) \right]^{0.5},$$
 (13)

where

= test statistic,

 B_{0p} = intercept of relation with current water quality.

= intercept of relation with future water quality,

= number of current measurements, m

= number of future measurements, and n

scale measure of "goodness of fit" from Tobit regression relation with current water quality.

The value of t is calculated using equation 13, then compared to the value of t_{crit} which is the value of the t distribution for df_{crit} degrees of freedom and a 0.05 level of significance. The value for df_{crit} is calculated using the equation

$$df_{crit} = (m + n - 3),$$
 (14)

where

= degrees of freedom,

= number of current measurements, and = number of future measurements.

If the value of t equaled or exceeded the value of t_{crit}, the two intercepts are considered to be different, and water quality of the future is considered to be different than the current water quality.

The amount of change that can be detected in the intercept of the equation with current water quality can be estimated. Equation 13 can be rearranged into equation 15, from which the amount of change in intercept with current water quality can be detected.

$$(B_{0p} - B_{0f}) \approx (t_{crit})^* \left[((m+n)/(m*n))^* (scale^2) \right]^{0.5},$$
 (15)

where

= intercept of relation with current water quality,

 B_{of} = intercept of relation with future water quality,

= value of t distribution for given degrees t_{crit} of freedom of complex model and a 0.05 level of significance,

= number of current measurements of water m quality,

= number of future measurements of water nquality, and

= measure of "goodness of fit" from Tobit scale regression relation with current water quality.

For a relation in which the value of the water-quality characteristic is untransformed, the change in intercept which can be identified is

$$delta = \left[(B_{0f} - B_{0p}) / (B_{0p}) \right] * 100, \tag{16}$$

where

delta = change in intercept with current water
quality, in percent;

 B_{0p} = intercept in relation with current water quality; and

 B_{0f} = intercept in relation with future water quality.

For a relation in which the value of the water-quality characteristic is expressed as a logarithm, the change in intercept which can be identified is

$$delta \approx (\exp(B_{0f} - B_{0p}) - 1) * 100,$$
 (17)

where

delta = change in intercept with current water

quality, in percent;

exp = exponent;

 B_{0p} = intercept in relation with current water

quality; and

 B_{of} = intercept in relation with future water quality.

As an alternative to the comparison of intercepts, the nonparametric seasonal rank sum test (described in Helsel and Hirsch, 1992, p. 118 and 348) can also be used to determine if the values of water-quality characteristics measured in the future are, as a group, different from the values measured during 2002-04. This method, which compares the relative magnitudes of the two sets of values, does not require the values to be normally distributed and can incorporate nondetect values. The variation of water quality with season is considered.

If only a small fraction of the current and future values of a water-quality characteristic are nondetects, the effects of season and streamflow can be incorporated into the rank sum test. A LOWESS curve between all values (current and future) of a water-quality characteristic and streamflow is created. Then a seasonal rank-sum test determines whether the residuals from the LOWESS curve for future measurements are different from the residuals for 2002-04 measurements.

Water Quality of Streams in and near the Delaware Water Gap National Recreation Area, 2002-04

Water-quality characteristics discussed in this report were selected for two reasons. Some characteristics (such as phosphorus and nitrogen species) were selected because values of these characteristics are expected to change with increasing urbanization and because the differences in current land use among the drainage basins may help explain the station-to-station differences in current values. Other characteristics (such as calcium) were selected because the station-to-station differences in values of these characteristics are related to differences in the geology of the basins.

Summary of Water-Quality Characteristics at Each Station

Summary statistics for the values of water-quality characteristics at each station are shown in appendix 3, and boxplots showing the distribution of values of most water-quality characteristics are shown in figure 4. Boxplots of dissolved ammonia and dissolved orthophosphate phosphorus are not shown because the greatest reporting levels of nondetect values for these two characteristics were greater than most of the detected concentrations; as a result, little information would be displayed. Boxplots of the concentrations of total nitrogen and dissolved nitrate plus nitrite are censored because, at Vancampens Brook (station 01440100), 28 percent of the concentrations of total nitrogen and 59 percent of the concentrations of dissolved nitrate plus nitrite are nondetects.

Only those concentrations of dissolved oxygen (as percent of saturation) measured during the months of July, August, and September were used in the analyses because station-to-station differences in concentrations, as well as variations with streamflow, were expected to be more apparent if only these values were considered. The higher temperatures during July through September, relative to temperatures in the other months, tend to increase the rate and effects of the chemical and biological processes that either increase or decrease the amount of oxygen dissolved in the water. In addition, the effects of the processes that add or remove oxygen from the water are more pronounced or evident during the relatively low streamflow during July through September than during periods of higher streamflow.

Results of Analyses of Blanks and Replicates

For most major ions and plant nutrients, the results from analyses of blanks indicate that effects of contamination from sample-collection and sample processing equipment on concentrations appear negligible (table 5). Dissolved orthophosphate phosphorus, total nitrogen, organic nitrogen plus ammo-

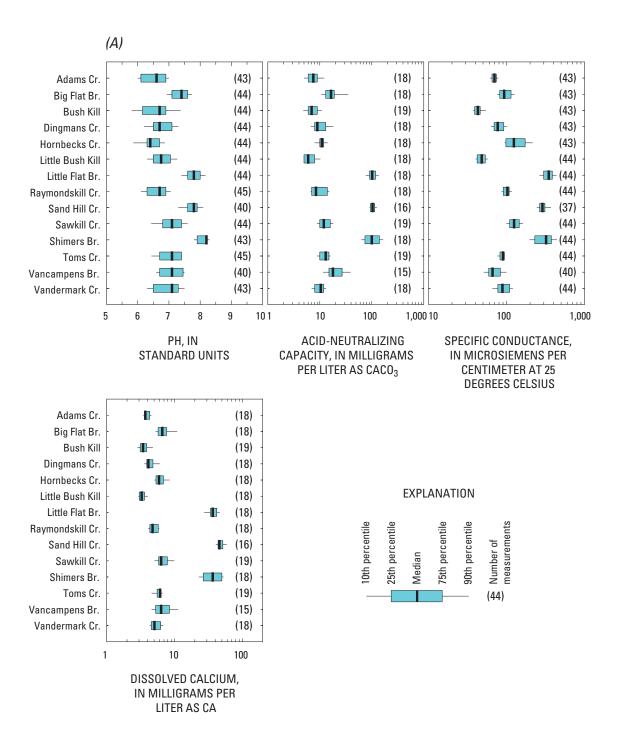


Figure 4. Distribution of values of (A) pH, acid-neutralizing capacity, specific conductance, dissolved calcium, (B) attenuation turbidity, total phosphorus, dissolved oxygen, total nitrogen, dissolved nitrate plus nitrite, and dissolved chloride at each water-quality monitoring station on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

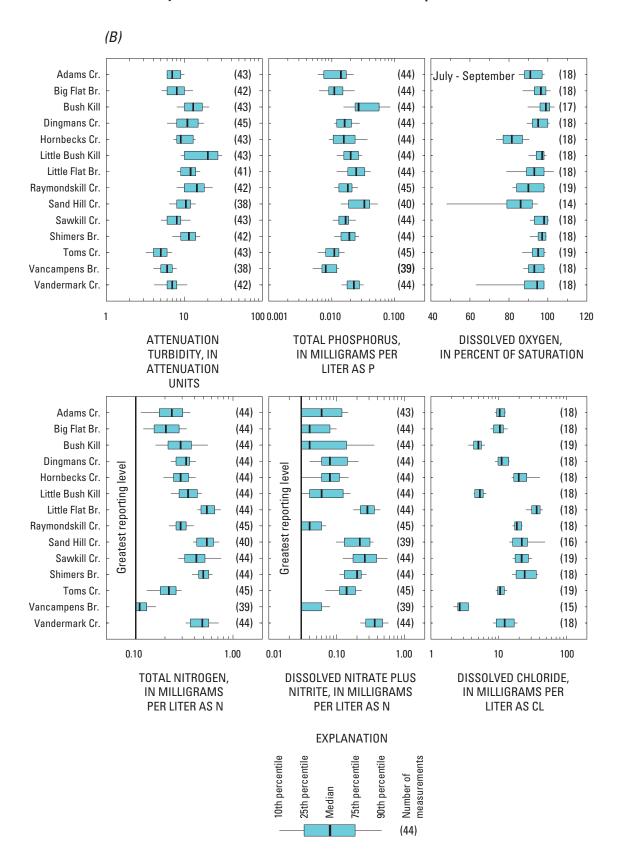


Figure 4. Distribution of values of (A) pH, acid-neutralizing capacity, specific conductance, dissolved calcium, (B) attenuation turbidity, total phosphorus, dissolved oxygen, total nitrogen, dissolved nitrate plus nitrite, and dissolved chloride at each water-quality monitoring station on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.—Continued

nia, dissolved nitrate plus nitrite, and dissolved chloride were not detected in any blank.

For total phosphorus and dissolved ammonia, the effects of contamination appear to exist but are small. Total phosphorus was detected in 2 of 14 blanks, and dissolved ammonia was detected in 1 of 14 blanks (table 5); detected concentrations of each were near the respective reporting levels.

Dissolved calcium was detected in three of seven blanks (table 5). The effect of this apparent contamination is negligible, however. The greatest detected concentration was 0.01 mg/L as Ca, two orders of magnitude lower than the smallest concentration in any environmental sample (about 2 mg/L as Ca).

Acid-neutralizing capacity was detected in four of seven blanks at concentrations near the reporting level (table 5). These detections could indicate contamination. However, they may also indicate that the laboratory reporting level during the study (1-2 mg/L as CaCO₃) was set too low and that concentrations at the reporting level, either in blanks or in samples, are not accurate. In 2005, the National Water Quality Laboratory raised the reporting limit of acid-neutralizing capacity to 5 mg/L as CaCO₃.

The summary of analyses of replicate/sample pairs (table 5) provides information for each characteristic on the differences between concentrations in replicates and associated samples in each replicate/sample pair. Each replicate constitutes a measurement of the water quality of a stream separated in time by a few minutes from the associated sample. Differences between concentrations in replicates and concentrations in the associated samples can be the result of changes in the concentrations in the stream between the time the sample was collected and the time the replicate was collected, but may also be the result of variations in methods of sample collection, processing, and (or) laboratory analysis.

The differences between concentrations in replicates and concentrations in associated samples provide a measure of the variability in concentrations in all samples collected as part of this sampling program that are due to these three factors. Variability in detection and variability in concentration were considered separately.

A measure of the variation in detection of each characteristic is indicated by examining the percentage of replicate/sample pairs with incongruous results. The characteristics with the largest percentage of replicate/sample pairs with incongruous results tend to be those with the smallest number of detected values (and, therefore, the smallest concentrations relative to the reporting limits) in the samples of the replicate/sample pairs (table 5). Dissolved ammonia is the characteristic with the most replicate/sample pairs with incongruous results (12 percent of pairs) and the fewest number of detections in the samples of the replicate/sample pairs (38 percent). In contrast, the characteristics with no incongruous results (total phosphorus, total nitrogen, dissolved calcium, and acid-neutralizing capacity) were detected in 100 percent of the samples of the replicate/sample pairs.

A measure of the variation in concentration for each characteristic is indicated by comparing concentrations in samples with concentrations in associated replicates. The variability in concentration appears to be affected by whether or not the concentration is for characteristics in dissolved form or dissolved plus particulate form. The greatest median differences (in terms of percentages of sample concentration) were for total phosphorus (8 percent), total nitrogen (7 percent), and total organic nitrogen plus ammonia (20 percent). All these plant nutrients are in dissolved plus particulate form. For plant nutrients in the dissolved form only, the median differences were less than 1 percent.

Two reasons are proposed to account for the greater variability in concentration, when characteristics are composed, at least partly, of particulate substances compared with those composed only of dissolved substances. First, the variation in concentration in the stream over short periods of time may be greater for particulate substances than for dissolved substances. Second, the method of creating subsamples from each composited sample (shaking and pouring from the compositing bottle) could have created subsamples with non-representative concentrations of particulate substances because the particles were not distributed evenly.

Variation of Water Quality with Basin Characteristics

It is apparent from the boxplots in figure 4 that values of a selected water-quality characteristic varied from stream to stream. For example, the median concentration of total phosphorus at the station on Sand Hill Creek (0.033 milligrams per liter as P) was four times the corresponding median concentration at the station on Vancampens Brook (0.008 milligrams per liter as P). This variation can be attributed to differences in land-use and geology of the drainage basins, as well as to the presence of wastewater discharges to the basins.

Results of correlations between median values of waterquality characteristics at the stations and the amounts of highintensity and low-intensity urban land, agricultural land, and wetlands are given in table 6. Relations for most water-quality characteristics were determined from medians from all 14 stations; relations for dissolved orthophosphate phosphorus and dissolved ammonia were determined from medians at 12 and 9 stations, respectively.

In all relations identified, median concentrations increased with increasing amounts of a specified land use (table 6). In no relations did median concentrations decrease with increasing amounts of any of the specified land uses.

For the two phosphorus and three nitrogen species, results of correlations between median concentrations of these plant nutrients and land use at all stations indicate that agricultural land appears to have a greater effect on water quality than either high-intensity or low-intensity urban land when all 14 stations are considered (table 6). Median concentrations of total nitrogen and dissolved nitrate plus nitrite increased with

Summary of results of analyses of blanks and replicates from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04. Table 5.

[mg/L, milligrams per liter; P, phosphorus; N, nitrogen; Cl, chloride; Ca, calcium; CaCO₃, calcium carbonate; n.a., not applicable; differences between concentrations in replicates and samples were set to positive values; differences between a detected concentration and a nondetected concentration were not determined]

			Blanks					Replicates	ates		
		Num-				Number of	Jo in the second	Differenc	es between co	Differences between concentrations in sample and replicate	n sample and
Water-quality characteristic	1	blanks	Greatest	Greatest	Num- her of	replicate/	replicate/	Median of	Median of differences	Maximum	Maximum of differences
uetermineu by laboratory analysis (units of concentra- tion)	Num- ber of blanks	in which charac- teristic was detec- ted	detected concentra- tion, in mg/L	reporting level of nondetect concentra- tions, in mg/L	repli- cate/ sample pairs	sample pairs) with detected concen- trations, in	sample pairs with only one detection, in percent	In units of concentra- tion	In percent of concen- tration in sample	In units of concen- tration	In percent of concentration in sample¹
					Plant nutrients	ents					
Total phosphorus (in mg/L as P)	14	7	0.003	0.002	26	100	0	0.001	∞	0.023	120
Dissolved orthophosphate-phosphorus (in mg/L as P)	13	0	n.a.	.01	26	54	∞	0	0	.01	100
Total nitrogen² (in mg/L as N)	В	0	n.a.	.03	12	100	0	.02	7	.05	19
Total organic nitrogen plus ammonia (in mg/L as N)	11	0	n.a.	.05	14	93	∞	.01	20	60.	09
Dissolved nitrate plus nitrite (in mg/L as N)	14	0	n.a.	.03	26	92	4	0	0	.00	25
Dissolved ammonia (in mg/L as N)	14	-	.005	.0075	26	38	12	0	0	.002	20
					Major ions	ns					
Dissolved chloride (in mg/L as Cl)	7	0	n.a.	.15	6	100	0	Т.	0.8	∞.	S
Dissolved calcium (in mg/L as Ca)	7	ϵ	.01	.005	6	100	0	.04	9.	∞.	က
Acid-neutralizing capacity (in mg/L as CaCO ₃)	7	4	7	2	6	100	0	0	0	0.	0

¹Calculated for those sets of environmental and replicate samples in which concentrations were detected.

² Determined from laboratory analysis.

Table 6. Results of correlation tests between median values of water-quality characteristics at each water-quality monitoring station and the land use in the drainage basin, for streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[+, median increased with increasing land use at a 0.05 level of significance; --, no significant correlation; <, less than]

Water-quality characteristic			rhether median value of the v ng land use in drainage basir	
	High-intensity urban land	Low-intensity urban land	Agricultural land	Wetlands
	Phys	ical characteristics		
Dissolved oxygen concentration (July-September)				
Specific conductance			+ (0.003)	
pH				
Attenuation turbidity				+ (<0.0001)
		Plant nutrients		
Total phosphorus				+ (0.028)
Dissolved orthophosphate phosphorus ¹				
Total nitrogen			+ (0.01)	
Dissolved nitrate plus nitrite			+ (0.027)	
Dissolved ammonia ²				
		Major ions		
Acid-neutralizing capacity				
Dissolved calcium				
Dissolved chloride	+ (0.025)		+ (0.0008)	

¹Only the uncensored medians at 12 stations were included in the analysis.

an increase in the area of agricultural land in the associated basin, but not with increases in the areas of either high-intensity or low-intensity urban land. Similar results for total nitrogen and dissolved nitrate plus nitrite can be attributed to the fact that total nitrogen is composed partly of nitrate plus nitrite; median concentrations of total nitrogen as a function of agricultural land use are shown in figure 5. No relations between either phosphorus species and agricultural land or high-intensity or low-intensity urban land were identified.

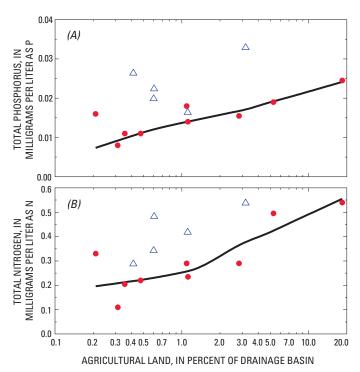
An examination of figure 5 indicates that the presence of a permitted wastewater facility in a basin appears to be a factor affecting median concentrations of total phosphorus and total nitrogen in the streams. For a specified amount of agri-

cultural land in a basin, concentrations at stations with permitted wastewater facilities in the basin appear to be greater than concentrations at stations without such facilities in the basin.

To provide clearer relations between water quality and land use without effects of permitted wastewater discharge, selected correlations indicated in table 6 were repeated without the 5 stations in basins with permitted wastewater facilities (table 2). The results of these correlations are described below.

When only the nine stations with no permitted wastewater facilities are considered, median concentrations of both total phosphorus and total nitrogen increased with an increase in the area of agricultural land use in the associated basin (fig. 5); each relation has a 0.01 level of significance. These

²Only the uncensored medians at 9 stations were included in the analysis.



EXPLANATION

- LOWESS smoothed curve through median concentrations at stations with no permitted wastewater facility
- Median concentration at a station with no permitted wastewater facility
- Median concentration at a station with a permitted wastewater facility

Figure 5. Median concentrations of (A) total phosphorus and (B) total nitrogen at water-quality monitoring stations as a function of agricultural land use and the presence of permitted wastewater facilities in associated drainage basins, for streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

patterns indicate that agricultural land use could be an important factor affecting the concentrations of both plant nutrients where no permitted wastewater discharges are present. The importance of agricultural land as a factor in streamwater quality has been discussed previously.

When all stations are considered, the area of wetlands in a basin does not appear to be a major factor related to median concentrations of phosphorus and nitrogen species. Of the two phosphorus and three nitrogen species, relations between median concentration and the area of wetlands in a basin were only identified for total phosphorus (table 6); median concentrations of total phosphorus increased with increasing area of wetlands (level of significance is 0.028).

When only the seven stations in basins with (a) no permitted wastewater facility and (b) agricultural land use accounting for less than 5 percent of the basin were consid-

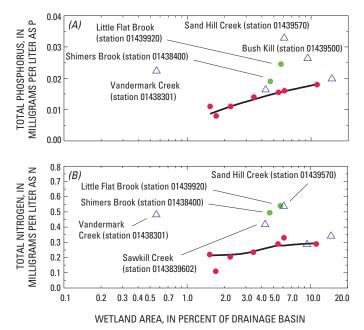
ered, median concentrations of both total phosphorus and total nitrogen increased with the increasing area of wetlands in the basin (fig. 6). The level of significance for the relation for total phosphorus is 0.003, and the relation for total nitrogen is 0.03. The two stations with more than 5 percent agricultural land within the basin, but with no permitted wastewater discharge (Shimers Brook and Little Flat Brook), were not included in the analysis to minimize the effects of agriculture on the relation between water quality and wetland area.

Little information is available in the literature about yields of phosphorus and nitrogen from wetlands. Wetlands are commonly considered as sinks for phosphorus and nitrogen that enter in streamflow (Mitsch and Gosselink, 1986, table 5-8). Nevertheless, Dillon and Molot (1997) report that the concentrations of phosphorus in streamflow leaving mostly forested basins in Ontario, Canada, increased with an increasing fraction of the basin composed of wetlands; they attributed this result to the complexation (facilitated by iron) of phosphorus with dissolved organic carbon. Whether the processes described for the forested basins in Ontario occur in the basins of streams tributary to the DWGNRA is unknown. No similar report comparing nitrogen yields from wetlands and forested basins could be found.

The small percentages of agricultural and wetland area in the drainage basin of Vancampens Brook (station 01440100) could be one reason why the median concentrations of total phosphorus, total nitrogen, and dissolved nitrate plus nitrite for this station were the smallest of those for all stations (fig. 4). The basin draining to this station has the second smallest area of agricultural land and wetlands of any basin considered.

From figure 6, the presence of a permitted wastewater facility that discharges into the drainage basin appears to affect the concentrations of total phosphorus and total nitrogen in streamflow leaving some, but not all, basins receiving such discharges. Median concentrations of total phosphorus at three of the five stations with permitted discharges—Vandermark Creek (station 01438301), Sand Hill Creek (station 01439570), and Bush Kill (station 01439500)—were noticeably greater than the median concentrations at stations with no permitted discharges and with less than 5 percent agricultural land in the basin. Also, from examination of the graph in figure 6, median concentrations of total nitrogen at three of the five stations with permitted discharges—Vandermark Creek (station 01438301), Sand Hill Creek (station 01439570), and Sawkill Creek (station 0143839602)—were noticeably greater than median concentrations at stations with no permitted discharges and with less than 5 percent agricultural land in the basin.

The high median concentrations of nutrients at stations with permitted wastewater discharge (figs. 5 and 6) could be due to the effects of wastewater discharged directly to the streams. Some of wastewater facilities are septic systems, however, and the high concentrations could be the result of ground water leaving the septic systems and discharging to streams. Lastly, the high concentrations also could reflect the



EXPLANATION

LOWESS smoothed line through median concentrations at waterquality monitoring stations with no permitted wastewater facility and agricultural land use is less than 5 percent of the basin area

MEDIAN CONCENTRATION AT A WATER-QUALITY MONITORING STATION

- Station with no permitted wastewater facility and agricultural land use is greater than or equal to 5 percent of the basin area
- Station with no permitted wastewater facility and agricultural land use is less than 5 percent of the basin area
- △ Station with permitted wastewater facility

Figure 6. Median concentrations of *(A)* total phosphorus and *(B)* total nitrogen at water-quality stations as a function of wetland area and the presence of permitted wastewater facilities in associated drainage basins, for streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

presence of urban land or other land use sometimes associated with large wastewater facilities.

The median concentrations for Vandermark Creek (station 01438301) and Sawkill Creek (station 0143839602) may also be affected by the on-lot septic systems in Milford, Pa. Milford, a borough of about 1,000 residents, is almost entirely served by on-lot septic systems (Lenny Schall, Pennsylvania Department of Environmental Protection, oral commun., 2007). The basins of both streams include parts of this borough.

The median attenuation turbidity in all 14 streams appears to be strongly related to the wetland area in the associated basin. Median values of attenuation turbidity increased with an increase in the area of wetlands in the associated basin (table 6; fig. 7); the level of significance of this relation is <0.0001. Given that most of the measurements of attenua-

tion turbidity were conducted under base-flow conditions, the values of attenuation turbidity are more likely to represent the true color of the streamflow due to dissolved materials and less likely to represent the particulate material in the water. Therefore, this relation appears to be between the color of the streams and the amount of wetland in the basins.

The correlation between attenuation turbidity and amount of wetlands in the drainage basin likely is due to the greater amount of dissolved organic material in streamflow leaving wetlands than is present in streamflow leaving other types of land use; the color of water in wetlands typically is directly related to the amount of dissolved organic material in the water. Hem (1985) notes that "color in natural water usually results from leaching of organic debris" and that "intensely colored waters occur in many environments where vegetation is plentiful, as in swamps and bogs." Concentrations of dissolved organic carbon (a measure of dissolved organic material) are greatest in surface water in wetland areas compared to that in other types of natural waters (Thurman, 1985).

Relations between color and (or) concentration of dissolved organic carbon in streams and lakes and the percentage of drainage basin composed of wetlands are reported by others. Gergel and others (1999) report concentrations of dissolved organic carbon in lakes and rivers in Wisconsin increased with an increase in the wetlands in the associated drainage basins. Dillon and Molot (1997) report that the concentrations of dissolved organic carbon and the color of streams increased with increasing wetland area in mostly forested basins in Ontario, Canada.

Median values of dissolved chloride concentrations and specific conductance increased with increasing area of agricultural land in the associated basins (table 6). The reasons for these two relations are unknown, but the correlations may be attributed to fertilizer use. The smallest median concentrations of dissolved chloride measured in Vancampens Brook (station 01440100) (fig. 4) likely are due, in part, to the small amount of agricultural land in the basin; this stream drains the basin with the second smallest amount of agricultural land.

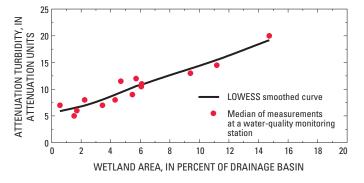


Figure 7. Median values of attenuation turbidity at waterquality monitoring stations as a function of wetland area in the associated drainage basins, for streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

When all stations are considered, median concentrations of dissolved chloride increased with increasing area of urban land use (table 6), a pattern likely affected by the presence of discharge from wastewater facilities. When only the nine stations in basins without permitted wastewater facilities are considered, however, there is no relation between urban land and median concentrations of dissolved chloride.

Median values of pH, specific conductance, dissolved calcium, and acid-neutralizing capacity in Little Flat Brook (station 01439920), Sand Hill Creek (station 01439570), and Shimers Brook (station 01438400) are the greatest of those from the 14 streams sampled (fig. 4). A comparison of a map of the drainage basins to a geologic map indicates that the basins associated with these three streams are underlain by carbonate rocks. The presence of carbonate rocks in the basins of Little Flat Brook and Shimers Brook in New Jersey was confirmed by an examination of the bedrock geologic map of northern New Jersey (Drake and others, 1996), but corresponding maps of the Sand Hill Creek basin were not readily available. The bedrock geologic map of northern New Jersey (Drake and others, 1996) does indicate that carbonate rocks likely underlie the Sand Hill Creek basin in Pennsylvania, however. Carbonate rocks are, generally, more soluble than most other types of rock; the presence of carbonate rock in a basin tends to raise the values of these water-quality characteristics in the stream draining the basin to levels that probably would not occur if the carbonate rocks were not present in the basin. ANC, a measure of the ability of streamwater to react with and neutralize acid, is a function of the amount of bicarbonate and carbonate in solution (Hem, 1985). Calcium is a major component of carbonate rocks.

Linear Equations Relating Water Quality, Streamflow, and Season

Linear equations relating values of water-quality characteristics, streamflow, and season (appendix 4) were developed as part of (1) the analyses relating water quality to streamflow and, separately, to season, and (2) detection of changes in water quality over time. An equation was developed if there were a sufficient number of detected values of a water-quality characteristic, if most of the residuals were normally distributed, and if the predicted values of water-quality characteristics equation appeared to match the measured values. The intercept and level of significance of the intercept, coefficients and levels of significance of the coefficients, and the value of scale in each equation are shown in appendix 4.

Variation of Water Quality with Streamflow and Season

Variations in values of a water-quality characteristic with streamflow and (or) season were examined because, in part, they describe how water quality varies over time. An understanding of how current water quality varies over time is needed in order to determine whether the water quality measured in the future is different from current water quality.

Variations of values of a water-quality characteristic with streamflow and (or) season also provide information on other factors that affect water quality. For instance, a decrease in concentrations of a characteristic with increasing streamflow could indicate that ground-water and (or) point-source discharges are important sources of that characteristic. In contrast, an increase in concentrations of a characteristic with increasing streamflow could indicate that material washed off the land surface during storms is an important source.

Most water-quality characteristics varied with increasing streamflow at most stations (table 7). Nine of the water-quality characteristics increased or decreased with increasing streamflow at nine or more stations.

Concentrations of dissolved orthophosphate phosphorus, dissolved nitrate plus nitrite, and dissolved ammonia decreased with increasing streamflow at more stations than concentrations increased with increasing streamflow. For example, concentrations of dissolved nitrate plus nitrite decreased with increasing streamflow at 9 stations and increased with increasing streamflow at 0 stations. These relations indicate that point-source discharges and (or) groundwater discharge (including septic-system discharge) are probable sources of these nutrients. An example graph showing decreasing concentrations of dissolved nitrate plus nitrite as a function of increasing streamflow is illustrated in figure 8. The relation between the concentrations and season shown in figure 8 is discussed below.

Concentrations of total phosphorus and total nitrogen were more likely to increase with increasing streamflow than to decrease (table 7). Concentrations of total phosphorus increased at five stations and decreased at one; concentrations of total nitrogen increased at seven stations and decreased at three. These increases could indicate that some of the phosphorus and nitrogen is associated with particulate material, which commonly increases in concentration with increases in streamflow. An example of concentrations of total phosphorus increasing with increasing streamflow is shown in figure 9; an example of concentrations of total nitrogen increasing with increasing streamflow is shown in figure 10.

The three stations at which concentrations of either total phosphorus or total nitrogen decreased with increasing streamflow—Bush Kill (station 01439500), Sawkill Creek (station 0143839602), and Vandermark Creek (station 01438301)—are all in basins with permitted wastewater facilities. The presence of permitted wastewater facilities may explain the decreasing concentrations with increasing streamflow if the permitted facilities discharged large amounts of phosphorus or nitrogen relative to other sources.

Values of attenuation turbidity increased with increasing streamflow at all stations (table 7); one example is given in figure 11. Increasing turbidity with streamflow is common and likely the result of particulate material washing off land surface during storms and the suspension of streambed material;

Identification of increases and decreases in values of selected water-quality characteristics with increasing streamflow, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[I, value increases with increasing streamflow; D, value decreases with increasing streamflow; --, values did not increase or decrease with increasing streamflow; n.d., not determined; Increases and decreases were identified with Tobit regression at a 0.05 level of significance; SC, specific conductance; ANC, acid-neutralizing capacity]

							Water-quality characteristic	/ characteris	tic				
Water-quality station	y station	Dis-	S	=	Atte- nua-	Total	Dissolved ortho-	Total	Dissolved	Dissol-	C	Dis-	Dis-
Short name	Station number	solved oxygen	36	E.	turbid- ity	pnos- phorus	pnospnate phospho- rus	nitrogen	nnrate plus nitrite	ved am- monia	ANC	solved	solved chloride
Adams Creek	01438754	I	D	D	I	n.d.	n.d.	1	D	1	D	D	
Big Flat Brook	01439830	1	D	D	Ι	Ι	n.d.	Ι	О	n.d.	D	D	D
Bush Kill	01439500	D	n.d.	D	I	D	О	D	D	D	D	D	D
Dingmans Creek	01438892	Ι	D	D	Ι	Ι	О	Ι	D	1	D	D	О
Hornbecks Creek	01439092	Ι	D	D	Ι	1	n.d.	Ι	1	D	О	D	D
Little Bush Kill	01439680	1	D	D	Ι	Ι	О	Ι	О	1	О	D	О
Little Flat Br	01439920	Ι	D	n.d.	Ι	n.d.	Ι	Ι	1	D	D	D	D
Raymondskill Cr	01438700	I	D	D	Ι	1	n.d.	Ι	1	D	D	D	1
Sand Hill Creek	01439570	I	D	Ι	Ι	1	1	1	1	1	1	D	D
Sawkill Creek	0143839602	1	D	D	Ι	Ι	1	О	О	n.d.	О	D	D
Shimers Brook	01438400	1	D	D	Ι	Ι	1	Ι	1	I	D	D	D
Toms Creek	01439400	1	D	1	Ι	1	О	1	О	n.d.	D	D	1
Vancampens Br	01440100	I	D	D	Ι	1	n.d.	1	О	n.d.	О	D	D
Vandermark Cr1	01438301	I	D	1	I	n.d.	1	D	D	n.d.	D	D	D
						Number	Number of stations with increases or decreases	h increases c	ır decreases				
	Increases	∞	0		14	5		7	0		0	0	0
	Decreases	1	13	10	0	1	4	3	6	4	13	14	11

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

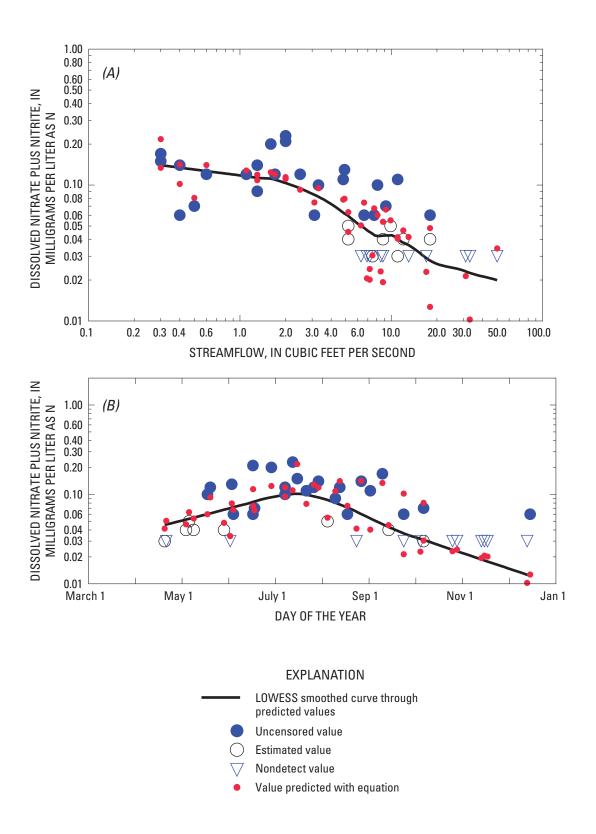


Figure 8. Concentration of dissolved nitrate plus nitrite as a function of (*A*) streamflow and (*B*) day of the year at the water-quality monitoring station (01438754) on Adams Creek below Long Meadow Brook near Edgemere, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

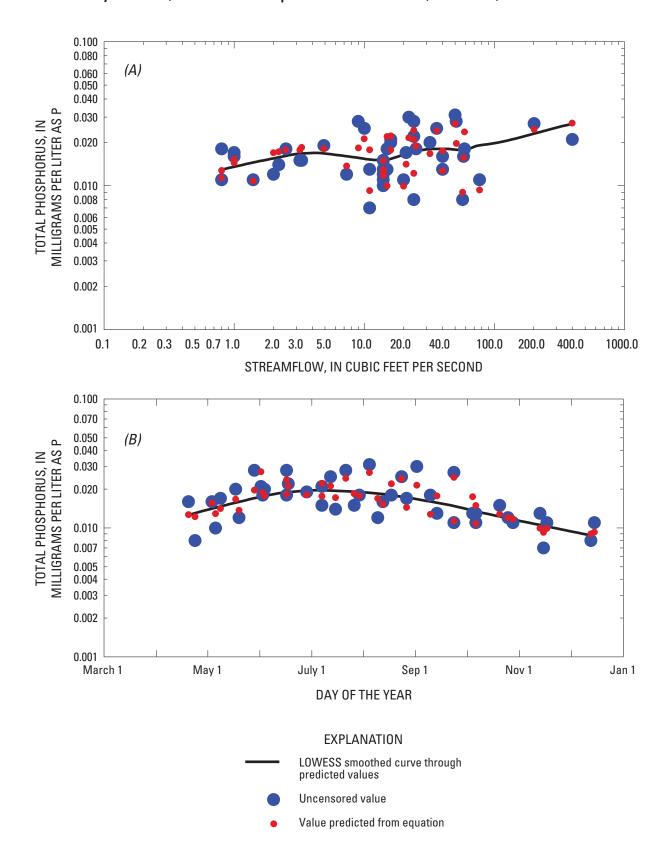


Figure 9. Total phosphorus as a function of (A) streamflow and (B) day of the year at the water-quality monitoring station (01438892) on Dingmans Creek above Dingmans Falls near Dingmans Ferry, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

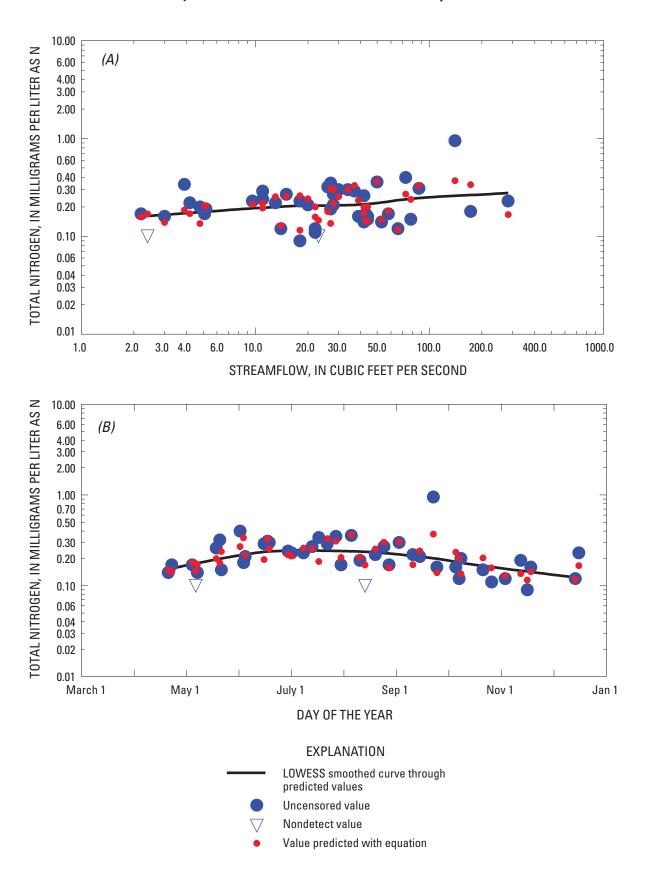


Figure 10. Total nitrogen concentration as a function of (A) streamflow and (B) day of the year at the water-quality monitoring station (01439830) on Big Flat Brook at Tuttles Corner, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

however, as previously discussed, the values of attenuation turbidity include the effects of true color as well as particles.

Values of acid-neutralizing capacity, dissolved calcium, dissolved chloride, and specific conductance at almost all stations decreased with increasing streamflow (table 7); one example is given in figure 12. These water-quality characteristics are measurements of major ions, and concentrations of major ions often decrease with increasing streamflow because the concentrations in precipitation are much lower than concentrations in ground water.

Values of pH decreased with increasing streamflow at 10 stations (table 7); the relation between pH and streamflow at the station (01439680) on Little Bush Kill at Bushkill, Pa., is shown in figure 13. This pattern is likely the result of the low pH of the precipitation in this area; during 2002-04, the weighted mean annual field pH of precipitation in this area ranged from 4 to 5 (National Atmospheric Deposition Program, 2006). As streamflow increases, the streamflow is more likely to have come from surface runoff (which has had little time to react chemically with soil and rock) than from groundwater discharge (which has had more time to react chemically with soil and rock). As a result, the pH during periods of high streamflow is more likely to be closer to the pH of precipitation than is the pH of streamflow during periods of low flow when the streamflow is made up entirely of ground-water and point-source discharges. In contrast, values of pH at the station (01439570) at Sand Hill Creek at Bushkill, Pa., increased with increasing streamflow; the reasons for this are unclear.

During July-September, dissolved oxygen concentrations increased with increasing streamflow at eight water-quality stations (table 7); as an example, the relation between dissolved oxygen concentration and streamflow at the station

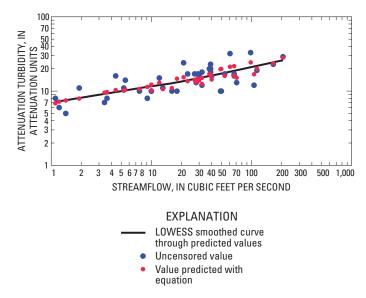
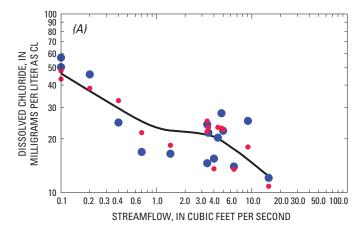


Figure 11. Attenuation turbidity as a function of streamflow at the water-quality monitoring station (01438700) on the Raymondskill Creek near Milford, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.



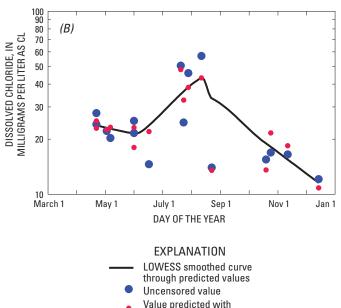


Figure 12. Dissolved chloride as a function of (A) streamflow and (B) day of the year at the water-quality monitoring station (01439570) on the Sand Hill Creek at Bushkill, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

equation

(01438700) on Raymondskill Creek near Milford, Pa., is shown in figure 14. In streams, there are processes adding oxygen (such as aeration and photosynthesis) and processes which are removing oxygen (such as benthic oxygen demand). These relations indicate that, at higher flows, the rate at which processes remove oxygen from the stream appears to be less than the rate at which processes add oxygen; whereas, at lower flows, the rate at which processes remove oxygen appears to exceed the rate at which processes add oxygen. Low dissolved oxygen concentrations in streamflow leaving wetlands may also be affecting the concentrations in streams during periods of low flow. In contrast, dissolved oxygen concentrations at the station (01439500) on Bush Kill near Shoemakers, Pa.,

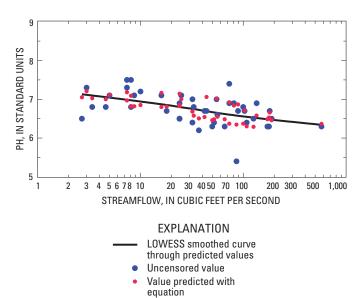


Figure 13. Values of pH as a function of streamflow at the water-quality monitoring station (01439680) on the Little Bush Kill at Bushkill, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J.,2002-04.

decreased with increasing streamflow, possibly as a result of supersaturation from photosynthesis at low streamflow.

Results of tests to determine whether values of a waterquality characteristic at a station varied with season are shown in table 8. For this analysis, season is represented by the Julian day of the year.

Due to the method of identification of seasonality, the variation of a water-quality characteristic with season is the result of the seasonal variation in conditions or processes other than streamflow. These other conditions or processes include (1) the loads of that water-quality characteristic, that are discharged to the stream, and (or) (2) instream biological, chemical, and (or) physical processes that affect water quality. The results of this analysis only identified whether or not values of water-quality characteristics varied with season; the analysis did not determine the seasons during which the values were greatest or smallest. For a given water-quality characteristic, seasonal relations at different stations may indicate different seasonal patterns at different stations. The relation between a water-quality characteristic and season does not necessarily represent conditions during January-March because no waterquality characteristics were measured during these months.

For 10 of the 11 water-quality characteristics analyzed, values varied seasonally at 7 or more stations (table 8). Values of dissolved oxygen were not tested for seasonality because only the measurements made during July-September were considered in this analysis.

Concentrations of nitrogen and phosphorus species varied with season at almost all stations for which calculations were made (table 8). Concentrations during the summer (July-September) often exceeded concentrations during the spring (April-June) and fall (October-December). Examples of this

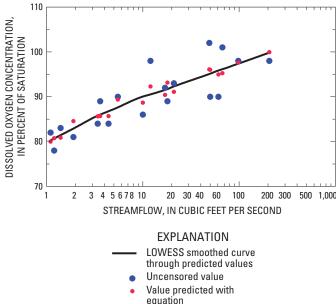


Figure 14. Dissolved oxygen concentration as a function of streamflow at the water-quality monitoring station (01438700) on the Raymondskill Creek near Milford, Pa., near the border of the Delaware Water Gap National Recreation Area, Pa. and N.J., July-September, 2002-04.

seasonal variation are shown in figures 8 to 10. As one example, concentrations of total nitrogen in Big Flat Brook (station 01439830) are between 0.1 and 0.2 milligrams per liter as N in the spring and fall, but increase to between 0.2 and 0.3 milligrams per liter as N during the summer. This pattern may indicate that sources of phosphorus and nitrogen were greater during summer than during spring or fall. This pattern would be consistent with the greater discharges from wastewater-treatment plants and septic systems in the study area during the summer that results from increased summer population. Another possibility is that this pattern is due to increased wash off of fertilizers from agricultural lands, although the greatest loads from fertilizer washing off agricultural land might occur during rainstorms immediately following a major application in the spring.

The concentration of dissolved chloride varied seasonally at nine stations (table 8). At some stations, the seasonal variation in concentrations of dissolved chloride followed the pattern as shown by nitrogen and phosphorus concentrations; concentrations during July-September were greater than those during other months. For example, the greatest concentrations of dissolved chloride at the Sand Hill Creek at Bushkill, Pa., station (01439570) were measured during July-September (fig. 12). This pattern may indicate that discharge from wastewater-treatment facilities (including septic systems) are important sources of chloride in some basins.

Identification of water-quality characteristics with values that varied with season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[S, value varied with season; --, value did not vary with season; n.d., not determined; Variation with season was identified with a nested F-test and Tobit regression at a 0.05 level of significance; SC, specific conductance; ANC, acid-neutralizing capacity]

						>	Water-quality characteristic	haracteristi	E)				
Water-quality station	ty station	Dis-		:	Attenua-	Total	Dissolved ortho-	Total	Dis- solved	Dissolved		Dis-	Dis-
Short name	Station number	oxygen	26	E.	tion turbidity	phos-	phosphate phospho- rus	gen	nitrate plus nitrite	ammonia	ANC	solved	solved chloride
Adams Creek	01438754	n.d.	S	S	S	n.d.	n.d.	S	S	S	S	S	1
Big Flat Brook	01439830	n.d.	1	S	S	S	n.d.	S	S	n.d.	S	1	1
Bush Kill	01439500	n.d.	n.d.	S	S	1	1	1	S	ŀ	S	1	S
Dingmans Creek	01438892	n.d.	S	S	S	S	S	S	S	S	S	S	S
Hornbecks Creek	01439092	n.d.	S	1	S	S	n.d.	S	S	S	S	S	S
Little Bush Kill	01439680	n.d.	S	S	S	S	S	S	S	S	S	S	S
Little Flat Br	01439920	n.d.	1	n.d.	S	n.d.	S	S	I	S	l	1	1
Raymondskill Cr	01438700	n.d.	S	S	S	S	n.d.	S	l	S	S	S	S
Sand Hill Creek	01439570	n.d.	1	S	S	S	S	S	l	S	I	!	S
Sawkill Creek	0143839602	n.d.	S	1	S	S	S	S	S	n.d.	S	1	S
Shimers Brook	01438400	n.d.	1	S	S	S	S	S	S	S	l	1	1
Toms Creek	01439400	n.d.	S	!	1	S	S	S	S	n.d.	S	!	S
Vancampens Br	01440100	n.d.	1	S	S	S	n.d.	S	S	n.d.	I	!	S
Vandermark Cr1	01438301	n.d.	1	1	1	n.d.	S	S	S	n.d.	S	!	1
					Nu	ımber of sta	Number of stations at which values varied with season	values varie	d with seasc	u			
		n.d.	7	6	12	10	8	13	11	8	10	5	6

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

Detection of Future Changes in Water Quality

In order for either method described in this report to accurately detect future changes in water quality, the techniques of field measurement, sample collection, and analysis in the future have to be the same as those used to measure water quality during 2002-04 (previously discussed). Otherwise, differences between current and future water quality identified by these or other methods could simply be an artifact of the differences in techniques.

Also, streamflow measurements at the times of the future sample collection will have to be available for at each station in order for the variation of water quality with streamflow to be considered in the analysis. For 13 of the stations, this means that instantaneous streamflow will need to be measured or estimated at the times of sample collection. For days on which samples are collected at the station (01438301) on Vandermark Creek at 4th Street at Milford, Pa., the record of daily streamflow for the station (01440000) Flat Brook near Flatbrookville, N.J., will need to be available.

Either of the two methods mentioned in this report—the seasonal rank sum test or the analysis of intercepts—can be used to identify future changes in water quality. The seasonal rank sum test can be used for values of any water-quality characteristic at any of the stations, whether or not equations could be developed that relate both current values and future values of the water-quality characteristic to streamflow and season. The analysis of intercepts can be used only if both equations could be developed for values of a water-quality characteristic at a station. Once an equation relating current values of a specified water-quality characteristic at a station to streamflow and season is developed, this method allows an estimate of the minimum change in current values of the water-quality characteristic at a station that can be detected.

The minimum detectable changes between current and future water-quality values are given in table 9 for those water-quality characteristics with equations relating current values of the characteristic at a station to streamflow and season (Appendix 4). To calculate these values, 10 measurements of future water quality and a 0.05 level of significance were assumed.

The minimum detectable changes in table 9 are expressed as the percentage of the intercept for the equations with current values of water quality. Each intercept represents an estimate of the current value of a specified water-quality characteristic at a specified station when streamflow was 1 ft³/s during a season in which the variation in water quality due to season was minimal. If the left-hand side of the equation is the logarithm of the water quality, the intercept represents the logarithm of the value of the water-quality characteristic under these conditions.

The minimum detectable changes varied among waterquality characteristics (table 9). For the water-quality characteristics composed of or related to major ions (pH, specific conductance, dissolved calcium, acid-neutralizing capacity, or dissolved chloride) minimum detectable changes were 1 to 9 percent of intercepts. For attenuation turbidity, total phosphorus, and total nitrogen, the minimum detectable changes were 3 to 12 percent; and for dissolved orthophosphate phosphorus, dissolved nitrate plus nitrite, and dissolved ammonia, 6 to 30 percent.

Examination of equation 15 indicates that the magnitudes of minimum detectable changes (table 9) are related to the number of current values of water-quality characteristics and to the value of scale (goodness of fit) of the linear relations, which are presented in Appendix 4. The minimum detectable change decreases with an increase in the number of values of a current water-quality characteristic and increases with an increase in the value of scale (a measure of the error in the fit of the equation to the measured values).

When considering the changes (between current and future water-quality values) that are detectable using this method, it is assumed that the relations between future water quality and streamflow or season are similar to those between current water quality and streamflow, and season. Given the uncertainty of the relations in the future, the actual detectable changes are likely to be larger than those changes calculated in this report.

Assuming that future methods used to measure water-quality characteristics will be similar to the methods used to measure water-quality characteristics during 2002-04, any detectable changes between current and future water-quality values then can be attributed to either changes in basin characteristics or, possibly, to changes in climate. Basin characteristics that can change and result in changes in water quality over time include land use, wastewater-discharge facilities, and any human activity that alters the streamflow.

Summary and Conclusions

Water samples were collected at stations on 14 streams within or entering the Delaware Water Gap National Recreation Area, Pa., and N.J., during 2002-04 in order to define water quality and provide a baseline from which to detect changes in water quality over time. Measurements of physical characteristics and analyses of samples for plant nutrients and major ions were made from April through December of each year, mostly under base-flow conditions. Summary statistics were generated for dissolved oxygen, specific conductance, pH, attenuation turbidity, total phosphorus, dissolved orthophosphate phosphorus, total nitrogen, dissolved nitrate plus nitrite, dissolved ammonia, acid-neutralizing capacity, dissolved chloride, and dissolved calcium. Boxplots were created for all water-quality characteristics except dissolved orthophosphate phosphorus and dissolved ammonia.

For a given water-quality characteristic, median values commonly varied among the 14 water-quality monitoring stations. For example, the median concentration of total phosphorus at the station on Sand Hill Creek (0.033 milligrams per liter as P) was four times the corresponding median concentra-

Minimum detectable differences between current (2002-04) and future water-quality values for stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04, assuming 10 future measurements. Table 9.

[Difference can be positive or negative; Positive differences given here; DO, dissolved oxygen; SC, specific conductance; TP, total phosphorus, DOP, dissolved orthophosphate; TN, total nitrogen, NO23, dissolved nitrate plus nitrite; DNH3, dissolved ammonia; ANC, acid-neutralizing capacity; Ca, dissolved calcium, Cl, dissolved chloride; n.d., could not be determined]

			Amount of c	lifference	Amount of difference in water-quality values that can be detected, in percent of intercept ($\mathtt{B_o}$) in Appendix 4	values that	can be dete	cted, in p	ercent of int	tercept (B ₀) i	n Append	ix 4	
Short name	Station number	D01)S	Н	Attenua- tion turbid- ity	Т	DOP	N	N023	DNH3	ANC	Ca	5
Adams Creek	01438754	3	2	3	9	n.d.	n.d.	9	12	16	3	2	4
Big Flat Brook	01439830	3	3	2	8	11	n.d.	6	17	n.d.	2	3	8
Bush Kill	01439500	3	n.d.	3	3	15	18	11	23	30	4	3	4
Dingmans Creek	01438892	2	2	3	S	5	6	4	11	13	3	2	3
Hornbecks Creek	01439092	3	3	3	7	11	n.d.	9	15	16	4	3	4
Little Bush Kill	01439680	3	3	2	7	9	9	5	14	18	4	2	3
Little Flat Br	01439920	18	4	n.d.	7	n.d.	15	5	10	12	3	3	5
Raymondskill Cr	01438700	3	2	3	7	9	n.d.	4	11	12	ς	3	3
Sand Hill Creek	01439570	S	3	2	7	10	111	5	13	17	4	2	9
Sawkill Creek	0143839602	2	2	3	8	10	8	9	9	n.d.	2	2	3
Shimers Brook	01438400	3	3	1	7	6	16	4	6	12	4	3	5
Toms Creek	01439400	4	3	3	8	12	9	6	10	n.d.	2	2	5
Vancampens Br	01440100	3	3	3	S	6	n.d.	10	20	n.d.	3	2	3
Vandermark Cr^2	01438301	14	4	8	12	n.d.	10	5	9	n.d.	2	3	6
	Maximum	18	4	3	12	12	18	11	23	30	S	3	6
	Minimum	2	2	_	3	5	9	4	9	12	2	2	3

¹Measurements July to September.

^{*}Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

tion at the station on Vancampens Brook (0.008 milligrams per liter as P).

Median values of water-quality characteristics at all 14 stations were correlated with land-use characteristics of the drainage basins to identify land uses that appear to affect water quality. Results indicate that median concentrations of total nitrogen and dissolved nitrate plus nitrite increased with an increase in the area of agricultural land in the associated basin (levels of significance are 0.01 and 0.027, respectively), but not with increases in the area of either high-intensity or low-intensity urban land. No relations were identified between median concentrations of other phosphorus and nitrogen species and the area of either agricultural, high-intensity urban, or low-intensity urban land.

Correlations between median values of phosphorus and nitrogen species and land use were repeated with only the nine stations without permitted wastewater facilities in order to reduce any influence of large wastewater discharges on the relations. Results indicate that agricultural practices could be an important factor affecting the concentrations of both total nitrogen and total phosphorus in these streams. Median concentrations of both total phosphorus and total nitrogen increased with an increase in the area of agricultural land in the drainage basins; the levels of significance are 0.01 for total phosphorus and 0.01 for total nitrogen.

Results of correlations may also indicate that the areas of wetlands in a basin could be a factor affecting the concentrations of total nitrogen and total phosphorus in these streams. When only the seven stations without permitted wastewater facilities and with less than 5 percent of the basin composed of agricultural land were considered, median concentrations of total phosphorus and total nitrogen increased with an increase in the amount of wetland area; the levels of significance are 0.003 for total phosphorus and 0.03 for total nitrogen.

The presence of a permitted wastewater facility that discharges into the drainage basin may also affect the concentrations of total phosphorus and total nitrogen in streamflow leaving some of the five basins with such facilities. When plotted as a function of wetland area, median concentrations of total phosphorus and (or) total nitrogen at four of the five stations with wastewater facilities in their drainage basins (Vandermark Creek, station 01438301; Sand Hill Creek, station, 01439570; Sawkill Creek, station 0143839602; and Bush Kill, station 01439500) were noticeably greater than the median concentrations at stations with no permitted facilities and with less than 5 percent agricultural land in the basin. The high median concentrations of nutrients at stations with permitted wastewater discharge could be due to the effects of wastewater discharged directly to the streams, the result of ground water leaving septic systems and discharging to the stream, or reflect the presence of urban land or other land use sometimes associated with large wastewater facilities.

The median attenuation turbidity in all 14 streams appears to be strongly related to the wetland area in the associated basin. Median values of attenuation turbidity increased with an increase in the area of wetlands in the associated

basin; the level of significance of this relation is <0.0001. Given that most of the measurements of attenuation turbidity were conducted under base-flow conditions, the values of attenuation turbidity are more likely to represent the true color of the streamflow due to dissolved materials and less likely to represent the particulate material in the water.

Median values of pH, specific conductance, dissolved calcium, and acid-neutralizing capacity (ANC) in Little Flat Brook (station 01439920), Sand Hill Creek (station 01439570), and Shimers Brook (station 01438400) are the greatest of those in the 14 streams sampled. A comparison of a map of the drainage basins to a geologic map indicates that the basins associated with these three streams are underlain or are likely underlain by carbonate rocks. The presence of carbonate rocks in the three basins would account for the greater values of pH, specific conductance, ANC, and dissolved calcium in these three streams than in the other streams

Linear equations between values of each water-quality characteristic, streamflow, and season at each station were developed from Tobit regression. Whether values of water quality changed with increasing streamflow and (or) with changes in season were determined partly from an examination of these equations.

Concentrations of dissolved orthophosphate phosphorus, dissolved nitrate plus nitrite, dissolved ammonia, and major ions decreased with increasing streamflow at more waterquality stations than concentrations increased with increasing streamflow. For example, concentrations of dissolved nitrate plus nitrite decreased with increasing streamflow at nine stations and increased with increasing streamflow at no stations. These relations likely indicate that point-source discharges and (or) ground-water discharge (including septic-system discharge) are important sources of these constituents.

Concentrations of total phosphorus, total nitrogen, and attenuation turbidity increased with increasing streamflow at more stations than concentrations decreased with increasing streamflow. Concentrations of total phosphorus increased with increasing streamflow at five stations and decreased with increasing streamflow at one station; the corresponding values for total nitrogen increased at seven stations and decreased at three. This pattern occurred because values of these nutrients are affected by particulate material in the streams, and particulate material often increases with increasing streamflow.

Most water-quality characteristics varied with season at most stations due to reasons other than the seasonal variation in streamflow. For 10 of the 11 water-quality characteristics analyzed, changes with season were identified for 7 or more stations. Due to the method of identification of seasonality, the variation of a water-quality characteristic with season is the result of the seasonal variation in conditions or processes other than streamflow.

Concentrations of nitrogen and phosphorus species varied with season for most water-quality monitoring stations; concentrations of nitrogen and phosphorus species varied with season at 10 and 13 stations, respectively. Concentrations during the summer (July-September) often exceeded con-

centrations during the spring (April-June) and fall (October-December). As one example, concentrations of total nitrogen in Big Flat Brook (station 01439830) are between 0.1 and 0.2 milligrams per liter as N in the spring and fall but increase to between 0.2 and 0.3 milligrams per liter as N during the summer. This pattern may indicate that sources during the summer were greater than sources during spring or fall, which would be consistent with the greater discharges from wastewater-treatment plants and septic systems during the summer that result from increased summer population.

A method to detect changes in future water quality was described. The method is based on the linear equations developed between values of water-quality characteristics, streamflow, and season. For each water-quality characteristic at each station, the equation with current water-quality values is compared to the equation which will be determined with future water quality. Changes in future water quality are identified by testing for differences in the intercepts of the two equations. The intercept represents an estimate of the current value of a water-quality characteristic measured at a station (1) with a streamflow of 1 cubic foot per second and (2) during a season in which the seasonal variation of water quality is minimal.

The method to detect changes in future water quality was selected, in part, because it allows for an estimate of the minimum change in current water quality that can be detected. Assuming 10 measurements will be made in the future, the minimum detectable changes in total phosphorus or total nitrogen at any of the stations are 4 to 12 percent of the intercepts in equations for current water quality.

These methods can be used to generate an overall assessment of the current and potential effects of land development on water quality in streams entering the park. On the basis of data collected during the study, managers can develop numeric regulatory standards for the water quality of streams and a cost effective strategy for future monitoring. Use of standard protocols will allow for comparison of the results from this study with results from studies in other regions of the country. Information on efficient monitoring designs and the effects of residential development are relevant to most of the parks which have rivers or streams flowing into them from outside park boundaries.

References Cited

- American Public Health Association, 1998, Standard methods for the examination of water and wastewater (20th ed.): Washington, D.C., p. 1-1 through 10-159 plus plates and index.
- DeLuca, M.J., Heckathorn, H.A., Lewis, J.M., Gray, B.J., Melvin, E.L., Riskin, M.L., and Liu, N.A., 2005, Water resources data for New Jersey-water year 2004, volume 3, Water-quality data: U.S Geological Survey Water-Data Report NJ-05-3, 680 p.

- DeLuca, M.J., Heckathorn, H.A., Lewis, J.M., Gray, B.J., and Feinson, L.S., 2006, Water resources data for New Jerseywater year 2005, volume 3, Water-quality data: U.S Geological Survey Water-Data Report NJ-04-3, 568 p.
- Dillon, P.J., and Molot, L.A., 1997, Effect of landscape form on export of dissolved organic carbon, iron, and phosphorus from forested stream catchments: Water Resources Research, v. 33, n. 11, p. 2591-2600.
- Drake, Jr., A.A., Volkert, R.A., Monteverde, D.H., Herman,
 G.C., Houghton, H.F., Parker, R.A., and Dalton, R.F., 1996,
 Bedrock Geologic Map of Northern New Jersey: U.S. Geological Survey Miscellaneous Investigations Series, Map I-2540-A, 2 sheets.
- Eastern Brook Trout Joint Venture, 2006, Eastern Brook Trout, Status and Trends: Trout Unlimited, 30 p., accessed on June 20, 2007 at http://www.easternbrooktrout.org/statusta.html
- Federal Interagency Sedimentation Project, undated, U.S. DH-81 sampler, accessed on August 16, 2005, at http://fisp.wes.army.mil/Catalog_Page_US_DH-81_Sampler.htm
- Fischer, J.M., Riva-Murray, Karen, Hickman, R.E., Chichester, D.C., Brightbill, R.A., Romanok, K.M., and Bilger, M.D., 2004, Water quality in the Delaware River Basin, Pennsylvania, New Jersey, New York, and Delaware, 1998-2001: U.S. Geological Survey Circular 1227, 38 p.
- Fishman, M.J., ed., 1993, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of inorganic and organic constituents in water and fluvial sediments: U.S. Geological Survey Open-File Report 93-125, 217 p.
- Fishman, M.J., and Friedman, L.C., 1989, Methods for determination of inorganic substances in water and fluvial sediments: U.S. Geological Survey Techniques of Water-Resources Investigations, book 5, chap. A1, 545 p.
- Gergel, S.E., Turner, M.G., and Kratz, T.K., 1999, Dissolved organic carbon as indicator of the scale of watershed influence on lakes and rivers: Ecological Applications, v. 9, no. 4, p 1377-1390.
- Heisig, Paul M., 2000, Effects of residential and agricultural land uses on the chemical quality of baseflow of small streams in the Croton watershed, southeastern New York: U.S. Geological Survey Water-Resources Investigations Report 99-4173, 15 p.
- Helsel, D.R., 2005, Nondetects and data analysis, statistics for censored environmental data: Hoboken, New Jersey, John Wiley & Sons, Inc., 250 p.
- Helsel, D.R., and Hirsch, R.M., 1992, Statistical methods in water resources: New York, Elsevier Science Publishing Company, Inc., 522 p.

- Hem, John D., 1985, Study and interpretation of the chemical characteristics of natural water: U.S. Geological Survey Water-Supply Paper 2254, 263 p., 3 pl.
- Hirsch, R.M., 1982, A comparison of four streamflow record extension techniques: Water Resources Research, v. 18, no. 4, p. 1081-1088.
- Judge, G.G., Griffiths, W.E., Hill, R.C., and Lee, T.-C., 1980, The theory and practice of econometrics: New York, John Wiley and Sons, 793 p.
- LaMotte Company, 1997, LaMotte Smart Colorimeter Operators Manual: Chestertown, Maryland, 50 p., accessed on February 22, 2007 at http://www.lamotte.com/pages/common/pdf/instruct/1911manu.pdf
- Lister, A., Riemann, R., Lister, T., and McWilliams, W., 2005,
 Northeastern regional forest fragmentation assessment
 Rationale, methods and comparisons with other studies, in Proceedings of the Fifth Annual Forest Inventory and Analysis Symposium, 2003 November 18-20, New Orleans, Louisiana: Washington, DC, U.S. Department of Agriculture Forest Service, General Technical Report WO-69, 222 p.
- Mathey, Sharon B., ed., 1998, National Water Information System (NWIS): U.S. Geological Survey Fact Sheet FS-027-98, accessed August 16, 2005, at http://water.usgs.gov/pubs/fs/FS-027-98/
- Mitsch, W.J., and Gosselink, J.G., 1986, Wetlands: New York, Van Nostrand Reinhold, 537 p.
- National Atmospheric Deposition Program (NRSP-3), 2006, Isopleth Maps: NADP Program Office, Illinois State Water Survey, accessed June 9, 2006, at http://nadp.sws.uiuc.edu/ isopleths/annualmaps.asp
- National Park Service, 2006, Delaware Water Gap National Recreation Area and Middle Delaware National Scenic River Water quality, accessed June 20, 2007, at http://www.nps.gov/dewa/naturescience/water-quality.htm
- Oblinger-Childress, C.J., Foreman, W.T., Connor, B.F., and Maloney, T.J., 1999, New reporting procedures based on long-term method detection levels and some considerations for interpretation of water-quality data provided by the U.S. Geological Survey National Water Quality Laboratory: U.S. Geological Survey Open-File Report 99-193, 19 p.
- Patton, C.J., and Kryskalla, J.R., 2003, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Evaluation of alkaline persulfate digestion as an alternative to Kjeldahl digestion for determination of total and dissolved nitrogen and phosphorus in water: U.S. Geological Survey Water-Resources Investigations Report 03-4174, 33 p.

- Patton, C.J., and Truitt, E.P., 2000, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of ammonium plus organic nitrogen by a Kjeldahl digestion method and an automated photometric finish that includes digest cleanup by gas diffusion: U.S. Geological Survey Open-File Report 00-170, 31 p.
- Rantz, S. E., and others, 1982, Measurement and Computation of Streamflow, volumes 1 and 2: U.S. Geological Survey Water-Supply Paper 2175, 631 p.
- SAS Institute, Inc., 1999a, The Corr Procedure: Cary, N.C., SAS Procedures Guide, Version 8, p. 273-312.
- SAS Institute, Inc., 1999b, The Lifereg Procedure: Cary, N.C., SAS/STAT User's Guide, Version 8, p. 1761-1796.
- Shelton, L. R., 1994, Field guide for collecting and processing stream-water samples for the National Water-Quality Assessment Program: U.S. Geological Survey Open-File Report 94-455, 42 p.
- Thurman, E.M., 1985, Organic chemistry of natural waters: Boston, Mass., Martinus Nijhoff/Dr W. Junk Publishers, 497 p.
- U.S. Census Bureau, 2006a, State and County QuickFacts, accessed on July 6, 2006, at http://quickfacts.census.gov/qfd/index.html
- U.S. Census Bureau, 2006b, TIGER, TIGER/Line, and TIGER-related products, accessed on July 13, 2006, at http://www.census.gov/geo/www/tiger/index.html
- U.S. Geological Survey, 1999, The quality of our Nation's waters nutrients and pesticides: U.S. Geological Survey Circular 1225, 82 p.
- U.S. Geological Survey, 2006, National Land Cover Data: Sioux Falls, South Dakota, USGS Land Cover Institute, accessed July 11, 2006, at http://landcover.usgs.gov/
- Zaugg, S.D., Smith, S.G., Schroeder, M.P., Barber, L.B., and Burkhardt, M.R., 2002, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of wastewater compounds by polystyrene-divinylbenzene solid-phase extraction and capillary-column gas chromatography/mass spectrometry: U.S. Geological Survey Water-Resources Investigations Report 01-4186, 37 p.

Appendix 1. Methods used to determine flow at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., at times of water-quality-sample collection, 2002-04

Values of streamflow were measured or estimated at the times of sampling at 13 of the 14 stream stations (table 1-1). Streamflow was not determined for samples collected at the water-quality monitoring station (01438301) on Vandermark Creek at 4th Street at Milford, Pa., because, on an initial survey, field personnel could not find a good spot to measure streamflow.

Three methods were used to measure most values of streamflow at the 13 stations (table 1-1). All methods are based on discharge measurements made at the stations using the methods described in Rantz and others (1982). At four stations, most values of streamflow were set equal to the streamflow measured by USGS personnel on the days the samples were collected. At the station on the Bush Kill at Shoemakers, Pa. (No. 01439500), stage (the elevation of the streamwater level above a datum) is continuously measured by the USGS, and continuous values of streamflow were calculated from the rating curve (the relation between stage and streamflow) for this station. Values of streamflow at the time of sample collection were taken from the record of streamflow.

At the remaining nine stations, values of streamflow for most samples were based on measurements of stage at the time of sample collection and from a rating curve for the individual station developed with streamflow measurements made during 2002-04. Rating curves for these stations were retrieved from the records of the USGS; an example of one rating curve is shown in figure 1-1.

For some samples, streamflow could not be determined by use of the methods described above. For these samples, streamflow was estimated using a MOVE1 relation (Hirsch, 1982) and the record of daily streamflow at 01440000 Flat Brook near Flatbrookville, NJ (fig. 1); an example of one relation is given in figure 1-2. For each station, the MOVE1 curve describes the relation between the measurements of streamflow at the station and the daily streamflow at Flat Brook (01440000) on the days of sample collection. Values of streamflow determined in this way were considered to be "estimated" values and given an "E" remark code.

Table 1-1. Method used to determine streamflow at the time of collection of most water samples at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Br, Brook; Cr, Creek; --, streamflow not determined]

Short name	Station number	Station name	Method used to determine streamflow for most samples at station
Adams Creek	01438754	Adams Creek below Long Meadow Brook near Edgemere, PA	Rating curve and measurement of stage at time of sample
Big Flat Brook	01439830	Big Flat Brook at Tuttles Corner, NJ	Measurement on day of sample collection
Bush Kill	01439500	Bush Kill at Shoemakers, PA	Rating curve and continuous record of stage
Dingmans Creek	01438892	Dingmans Creek above Dingmans Falls near Dingmans Ferry, PA	Rating curve and measurement of stage at time of sample
Hornbecks Creek	01439092	Hornbecks Creek at Emery Road near Dingmans Ferry, PA	Rating curve and measurement of stage at time of sample
Little Bush Kill	01439680	Little Bush Kill at Bushkill, PA	Rating curve and measurement of stage at time of sample
Little Flat Br	01439920	Little Flat Brook at Peters Valley, NJ	Measurement on day of sample collection
Raymondskill Cr	01438700	Raymondskill Creek near Milford, PA	Rating curve and measurement of stage at time of sample
Sand Hill Creek	01439570	Sand Hill Creek at Bushkill, PA	Rating curve and measurement of stage at time of sample
Sawkill Creek	0143839602	Sawkill Creek 1000 ft above mouth at Milford, PA	Rating curve and measurement of stage at time of sample
Shimers Brook	01438400	Shimers Brook near Montague, NJ	Measurement on day of sample collection
Toms Creek	01439400	Toms Creek at Egypt Mills, PA	Rating curve and measurement of stage at time of sample
Vancampens Br	01440100	Vancampens Brook near Millbrook , NJ	Measurement on day of sample collection
Vandermark Cr	01438301	Vandermark Creek at 4th Street at Milford, PA	

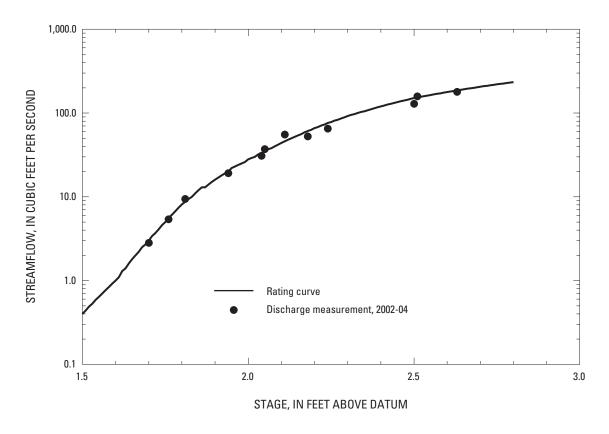


Figure 1-1. Rating curve and streamflow measurements of the Little Bush Kill at Bushkill, Pa. (station 01439680), 2002-04.

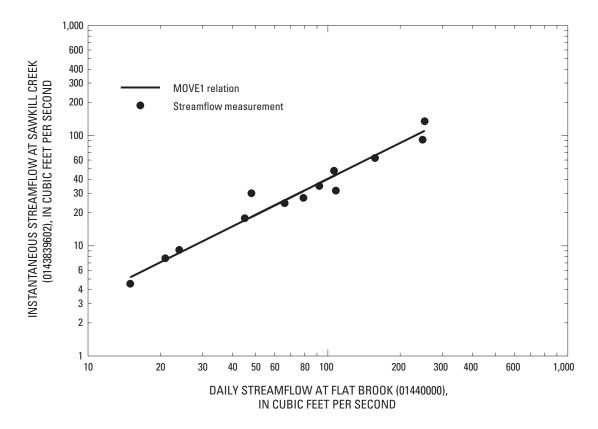


Figure 1-2. Relation between instantaneous streamflow of Sawkill Creek 1000 feet above mouth at Milford, Pa. (station 0143839602), and daily streamflow of Flat Brook at Flatbrookville, N.J. (station 01440000), 2002-04.

Appendix 2. A brief description of Tobit regression

Tobit regression (described in Judge and others, 1980) was used to develop all linear relations discussed in this report. All the relations are of the following form:

$$y_i^* = \sum B_i x_{ii} + e_i$$
 (2-1)

where

 y_i^* = value of experimental variable y in observation i (nondetect values are set equal to the reporting level),

 B_j = coefficient for explanatory variable j, x_{ii} = value of explanatory variable j in

observation i, and

 e_i = residual for observation i.

Tobit regression uses maximum likelihood estimation and iterative solutions to maximize the likelihood function in equation 2-2. The calculations, conducted with the Lifereg procedure (SAS Institute, Inc., 1999b), include estimating values for the residuals (e_i) as well as for the coefficients (B_j) in equation 2-1. Residuals are assumed to have a normal distribution and constant variance across the range of predicted values.

$$L = \prod \left[p(e_i)^{\delta_i} \times F(e_i)^{I - \delta_i} \right]$$
 (2-2)

where

L = likelihood,

 $p(e_i)$ = the probability density function of the residuals,

 $F(e_i)$ = the cumulative distribution function of the residuals,

 e_i = residual for observation i, and

 $\delta_i = 0$ for an observation *i* with a nondetect value of *y*, or

= 1 for an observation i with a detected value of y.

The goodness of fit of the relation to the measured values is indicated by the value of "scale." Scale is equivalent to mean square error of least-squares regression.

The Tobit relation may be strongly affected by outliers (Helsel and Hirsch, 1992). How well the values fit the linear equation and whether the residuals have constant variance and normality were evaluated as part of this report.

References Cited

Helsel, D.R., and Hirsch, R.M., 1992, Statistical methods in water resources: New York, Elsevier Science Publishing Company, Inc., 522 p.

Judge, G.G., Griffiths, W.E., Hill, R.C., and Lee, T.-C., 1980, The theory and practice of econometrics: New York, John Wiley and Sons, 793 p.

SAS Institute, Inc., 1999b, The Lifereg Procedure: Cary, N.C., SAS/STAT User's Guide, Version 8, p. 1761-1796.

Appendix 3. Summary statistics for water-quality characteristics in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

Index	
Water-quality characteristic	Table
Physical characteristics	
Dissolved oxygen	3-1
Specific conductance	3-2
pH	3-3
Attenuation turbidity	3-4
Plant nutrients	
Total phosphorus	3-5
Dissolved orthophosphate phosphorus	3-6
Total nitrogen	3-7
Dissolved nitrate plus nitrite	3-8
Dissolved ammonia	3-9
Major ions	
Acid-neutralizing capacity	3-10
Dissolved calcium	3-11
Dissolved chloride	3-12

Table 3-1. Summary statistics for dissolved oxygen concentrations in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., July-September, 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in percent of saturation; n.a., not applicable]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are nonde- tect values	Greatest reporting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	18	0	0	n.a.	83	91	99
Big Flat Brook	01439830	18	0	0	n.a.	84	97	104
Bush Kill	01439500	17	0	0	n.a.	87	99	104
Dingmans Creek	01438892	18	0	0	n.a.	83	95	103
Hornbecks Creek	01439092	18	0	0	n.a.	71	82	95
Little Bush Kill	01439680	18	0	0	n.a.	85	97	100
Little Flat Br	01439920	18	0	0	n.a.	77	93	127
Raymondskill Cr	01438700	19	0	0	n.a.	78	90	102
Sand Hill Creek	01439570	14	0	0	n.a.	39	86	97
Sawkill Creek	0143839602	18	0	0	n.a.	88	98	102
Shimers Brook	01438400	18	0	0	n.a.	90	97	107
Toms Creek	01439400	19	0	0	n.a.	80	95	105
Vancampens Br	01440100	18	0	0	n.a.	85	93	100
Vandermark Cr	01438301	18	0	0	n.a.	34	95	101

Table 3-2. Summary statistics for specific conductance in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in microsiemens per centimeter at 25 degrees Celsius; n.a., not applicable]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are nondetect values	Greatest reporting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	43	0	0	n.a.	59	70	83
Big Flat Brook	01439830	43	0	0	n.a.	53	93	147
Bush Kill	01439500	43	0	0	n.a.	30	43	68
Dingmans Creek	01438892	43	0	0	n.a.	59	77	110
Hornbecks Creek	01439092	43	0	0	n.a.	86	124	261
Little Bush Kill	01439680	44	0	0	n.a.	14	48	70
Little Flat Br	01439920	44	0	0	n.a.	213	347	485
Raymondskill Cr	01438700	44	0	0	n.a.	79	103	124
Sand Hill Creek	01439570	37	0	0	n.a.	155	288	460
Sawkill Creek	0143839602	44	0	0	n.a.	57	125	197
Shimers Brook	01438400	44	0	0	n.a.	175	321	511
Toms Creek	01439400	44	0	0	n.a.	52	90	100
Vancampens Br	01440100	40	0	0	n.a.	43	66	116
Vandermark Cr	01438301	44	0	0	n.a.	49	89	146

Table 3-3. Summary statistics for pH in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in standard units; n.a., not applicable]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are nondetect values	Greatest reporting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	43	0	0	n.a.	5.9	6.6	7.2
Big Flat Brook	01439830	44	0	0	n.a.	6.2	7.4	8.2
Bush Kill	01439500	44	0	0	n.a.	5.3	6.7	7.7
Dingmans Creek	01438892	44	0	0	n.a.	6	6.7	7.6
Hornbecks Creek	01439092	44	0	0	n.a.	5.2	6.4	7.3
Little Bush Kill	01439680	44	0	0	n.a.	5.4	6.8	7.5
Little Flat Br	01439920	44	0	0	n.a.	7.3	7.8	8.6
Raymondskill Cr	01438700	45	0	0	n.a.	5.9	6.7	7.2
Sand Hill Creek	01439570	40	0	0	n.a.	6.8	7.8	8.6
Sawkill Creek	0143839602	44	0	0	n.a.	5.7	7.1	7.7
Shimers Brook	01438400	43	0	0	n.a.	7.5	8.2	8.6
Toms Creek	01439400	45	0	0	n.a.	6.2	7.1	7.6
Vancampens Br	01440100	40	0	0	n.a.	6.1	7.1	7.9
Vandermark Cr	01438301	43	0	0	n.a.	6.1	7.1	7.8

Table 3-4. Summary statistics for attenuation turbidity in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in attenuation units; n.a., not applicable]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are nonde- tect values	Greatest reporting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	43	0	0	n.a.	5	7	17
Big Flat Brook	01439830	42	0	0	n.a.	4	8	66
Bush Kill	01439500	43	0	0	n.a.	6	13	26
Dingmans Creek	01438892	43	0	0	n.a.	2	11	24
Hornbecks Creek	01439092	43	0	0	n.a.	6	9	21
Little Bush Kill	01439680	43	0	0	n.a.	7	20	45
Little Flat Br	01439920	41	0	0	n.a.	7	12	53
Raymondskill Cr	01438700	42	0	0	n.a.	5	15	33
Sand Hill Creek	01439570	38	0	0	n.a.	4	11	25
Sawkill Creek	0143839602	43	0	0	n.a.	2	8	69
Shimers Brook	01438400	42	0	0	n.a.	3	12	41
Toms Creek	01439400	43	0	0	n.a.	3	5	27
Vancampens Br	01440100	38	0	0	n.a.	3	6	9
Vandermark Cr	01438301	42	0	0	n.a.	2	7	64

Table 3-5. Summary statistics for total phosphorus in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in milligrams per liter as P; E, estimated; n.a., not applicable]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are non- detect values	Greatest re- porting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	44	0	0	n.a.	E 0.004	0.014	0.073
Big Flat Brook	01439830	44	0	0	n.a.	.005	.011	.182
Bush Kill	01439500	44	0	0	n.a.	.011	.027	.21
Dingmans Creek	01438892	44	0	0	n.a.	.007	.016	.031
Hornbecks Creek	01439092	44	0	0	n.a.	.006	.016	.084
Little Bush Kill	01439680	44	0	0	n.a.	.011	.02	.044
Little Flat Br	01439920	44	0	0	n.a.	.007	.025	.176
Raymondskill Cr	01438700	45	0	0	n.a.	.008	.018	.053
Sand Hill Creek	01439570	40	0	0	n.a.	.01	.033	.168
Sawkill Creek	0143839602	44	0	0	n.a.	.008	.017	.189
Shimers Brook	01438400	44	0	0	n.a.	.009	.019	.102
Toms Creek	01439400	45	0	0	n.a.	E.003	.011	.069
Vancampens Br	01440100	39	0	0	n.a.	E.003	.008	.016
Vandermark Cr	01438301	44	0	0	n.a.	.01	.023	.084

Table 3-6. Summary statistics for dissolved orthophosphate phosphorus in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in milligrams per liter as P; <, less than; E, estimated]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are nondetect values	Greatest re- porting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	43	33	77	0.02	< 0.003	0.002	E 0.01
Big Flat Brook	01439830	43	40	93	.01	< .003	< .01	E .004
Bush Kill	01439500	44	11	25	.01	< .003	.01	.21
Dingmans Creek	01438892	44	30	68	.02	< .003	.003	E .02
Hornbecks Creek	01439092	44	40	91	.02	< .003	< .02	E .004
Little Bush Kill	01439680	44	22	50	.01	< .003	.01	.013
Little Flat Br	01439920	44	18	41	.01	< .003	.01	.04
Raymondskill Cr	01438700	45	34	76	.02	< .003	.002	E .01
Sand Hill Creek	01439570	39	9	23	.01	< .003	.01	.04
Sawkill Creek	0143839602	44	16	36	.01	< .003	.01	.02
Shimers Brook	01438400	44	28	64	.01	< .003	.003	.02
Toms Creek	01439400	45	22	49	.01	< .003	.01	E .01
Vancampens Br	01440100	39	29	74	.01	< .003	.003	E .01
Vandermark Cr	01438301	44	3	7	.01	< .01	.012	.04

Table 3-7. Summary statistics for measurements of total nitrogen at stations on streams in and near the Delaware Water Gap National Recreation Area, 2002-2004.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in milligrams per liter as N; <, less than]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements which are non- detect values	Greatest reporting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	44	0	0	n.a.	0.09	0.24	0.53
Big Flat Brook	01439830	44	2	5	0.1	< .1	.21	.95
Bush Kill	01439500	44	0	0	n.a.	.12	.29	1.2
Dingmans Creek	01438892	44	0	0	n.a.	.18	.33	.51
Hornbecks Creek	01439092	44	0	0	n.a.	.16	.29	5.6
Little Bush Kill	01439680	44	0	0	n.a.	.18	.35	.61
Little Flat Br	01439920	44	0	0	n.a.	.36	.54	1
Raymondskill Cr	01438700	45	0	0	n.a.	.17	.29	.52
Sand Hill Creek	01439570	40	0	0	n.a.	.33	.54	1.2
Sawkill Creek	0143839602	44	0	0	n.a.	.19	.42	1
Shimers Brook	01438400	44	0	0	n.a.	.28	.5	.87
Toms Creek	01439400	45	0	0	n.a.	.1	.22	.49
Vancampens Br	01440100	39	11	28	.1	< .03	.11	1.4
Vandermark Cr	01438301	44	0	0	n.a.	.28	.49	1.3

Table 3-8. Summary statistics for dissolved nitrate plus nitrite in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in milligrams per liter as N; n.a., not applicable; <, less than; E, estimated]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are non-detect values	Greatest reporting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	43	11	26	0.03	< 0.03	0.06	0.23
Big Flat Brook	01439830	44	17	39	.03	< .025	.04	.25
Bush Kill	01439500	44	17	39	.03	< .025	.04	1
Dingmans Creek	01438892	44	2	5	.03	< .03	.08	.32
Hornbecks Creek	01439092	44	3	7	.03	< .025	.08	5.35
Little Bush Kill	01439680	44	7	16	.03	< .025	.06	.25
Little Flat Br	01439920	44	0	0	n.a.	.07	.29	.5
Raymondskill Cr	01438700	45	10	22	.03	< .025	.04	.09
Sand Hill Creek	01439570	39	0	0	n.a.	.07	.22	.49
Sawkill Creek	0143839602	44	0	0	n.a.	.07	.26	.88
Shimers Brook	01438400	44	0	0	n.a.	.08	.2	.34
Toms Creek	01439400	45	0	0	n.a.	E .04	.14	.33
Vancampens Br	01440100	39	23	59	.03	< .025	.03	.13
Vandermark Cr	01438301	44	1	2	.025	< .025	.37	1.3

Table 3-9. Summary statistics for dissolved ammonia in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in milligrams per liter as N; <, less than; E, estimated]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are nonde- tect values	Greatest reporting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	43	29	67	0.0075	< 0.005	0.004	0.03
Big Flat Brook	01439830	44	40	91	.0075	< .005	< .0075	.018
Bush Kill	01439500	44	31	70	.0075	< .005	.002	.151
Dingmans Creek	01438892	44	30	68	.0075	< .005	.005	.029
Hornbecks Creek	01439092	44	17	39	.0075	< .005	.008	.064
Little Bush Kill	01439680	44	28	64	.0075	< .005	.005	.084
Little Flat Br	01439920	44	11	25	.0075	< .005	.008	.024
Raymondskill Cr	01438700	45	18	40	.0075	< .005	.008	.03
Sand Hill Creek	01439570	39	16	41	.0075	< .005	.008	.047
Sawkill Creek	0143839602	44	37	84	.0075	< .005	< .0075	E.009
Shimers Brook	01438400	44	29	66	.0075	< .005	.005	.025
Toms Creek	01439400	45	43	96	.02	< .005	< .02	E .005
Vancampens Br	01440100	38	37	97	.0075	< .005	< .0075	E.005
Vandermark Cr	01438301	44	39	89	.0075	< .005	< .0075	E .01

Table 3-10. Summary statistics for acid-neutralizing capacity in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in milligrams per liter as CaCO₃; n.a., not applicable; E, estimated]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are nonde- tect values	Greatest re- porting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	18	0	0	n.a.	5	8	E 14
Big Flat Brook	01439830	18	0	0	n.a.	9	17	39
Bush Kill	01439500	19	0	0	n.a.	4	7	15
Dingmans Creek	01438892	18	0	0	n.a.	6	9	E 23
Hornbecks Creek	01439092	18	0	0	n.a.	7	11	18
Little Bush Kill	01439680	18	0	0	n.a.	4	6	12
Little Flat Br	01439920	18	0	0	n.a.	64	102	E 158
Raymondskill Cr	01438700	18	0	0	n.a.	5	9	16
Sand Hill Creek	01439570	16	0	0	n.a.	69	104	139
Sawkill Creek	0143839602	19	0	0	n.a.	8	12	E 20
Shimers Brook	01438400	18	0	0	n.a.	56	101	E 172
Toms Creek	01439400	19	0	0	n.a.	8	13	E 19
Vancampens Br	01440100	15	0	0	n.a.	10	18	E 44
Vandermark Cr	01438301	18	0	0	n.a.	6	11	E 15

Table 3-11. Summary statistics dissolved calcium in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in milligrams per liter as Ca; n.a., not applicable]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are non- detect values	Greatest re- porting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	18	0	0	n.a.	2.97	3.73	5.2
Big Flat Brook	01439830	18	0	0	n.a.	3.99	6.6	12.3
Bush Kill	01439500	19	0	0	n.a.	2.59	3.44	5.97
Dingmans Creek	01438892	18	0	0	n.a.	3.32	4.13	6.96
Hornbecks Creek	01439092	18	0	0	n.a.	4.63	5.94	10.1
Little Bush Kill	01439680	18	0	0	n.a.	2.39	3.29	4.53
Little Flat Br	01439920	18	0	0	n.a.	23.1	36.75	53.3
Raymondskill Cr	01438700	18	0	0	n.a.	3.31	4.74	6.22
Sand Hill Creek	01439570	16	0	0	n.a.	32.5	46	65
Sawkill Creek	0143839602	19	0	0	n.a.	4.45	6.37	11.2
Shimers Brook	01438400	18	0	0	n.a.	20.4	36.7	55.7
Toms Creek	01439400	19	0	0	n.a.	4.53	6.17	6.9
Vancampens Br	01440100	15	0	0	n.a.	4.05	6.36	12.7
Vandermark Cr	01438301	18	0	0	n.a.	3.91	5.07	8.99

Table 3-12. Summary statistics for dissolved chloride in samples from water-quality monitoring stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Stations are listed in order of short name; greatest reporting level of nondetect values, minimum, median, and maximum are in milligrams per liter as Cl; n.a., not applicable]

Short name	Station number	Number of measure- ments	Number of nondetect values	Percentage of measurements that are non- detect values	Greatest re- porting level of nondetect values	Minimum	Median	Maximum
Adams Creek	01438754	18	0	0	n.a.	7.69	10.35	14.9
Big Flat Brook	01439830	18	0	0	n.a.	4.36	10.4	16.7
Bush Kill	01439500	19	0	0	n.a.	2.63	5.01	8.84
Dingmans Creek	01438892	18	0	0	n.a.	8.49	11.05	15.5
Hornbecks Creek	01439092	18	0	0	n.a.	12.9	19.7	53.6
Little Bush Kill	01439680	18	0	0	n.a.	3.55	5.27	8.04
Little Flat Br	01439920	18	0	0	n.a.	22.7	36.25	45.9
Raymondskill Cr	01438700	18	0	0	n.a.	13.7	18.55	26.4
Sand Hill Creek	01439570	16	0	0	n.a.	12.1	21.9	57
Sawkill Creek	0143839602	19	0	0	n.a.	12.4	21.8	34.8
Shimers Brook	01438400	18	0	0	n.a.	14.2	24.05	40.4
Toms Creek	01439400	19	0	0	n.a.	4.76	10.5	13.7
Vancampens Br	01440100	15	0	0	n.a.	1.97	2.66	3.68
Vandermark Cr	01438301	18	0	0	n.a.	7.31	12.2	25.8

Appendix 4. Equations relating water quality, streamflow, and season at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04

The equations relating water quality, streamflow, and season (listed in the index) are the basis for determining (a) relations between water quality and streamflow, and relations between water quality and season, and (b) changes in water quality over time. Intercepts and coefficients for streamflow and season are included. Also included are the levels of significance for each intercept and coefficient. All coefficients and the intercept remain in the equation regardless of the level of significance. Relations were determined with Tobit regression.

Relations for a water-quality characteristic at a station are not presented if certain requirements of the data were not met. These requirements are discussed in the text.

Index

Water-quality characteristic	Table
Physical characteristics	
Dissolved oxygen (July-September)	4-1
Specific conductance	4-2
pH	4-3
Attenuation turbidity	4-4
Plant nutrients	
Total phosphorus	4-5
Dissolved orthophosphate phosphorus	4-6
Total nitrogen	4-7
Dissolved nitrate plus nitrite	4-8
Dissolved ammonia	4-9
Major ions	
Acid-neutralizing capacity	4-10
Dissolved chloride	4-11
Dissolved calcium	4-12

4.2235 3.7034 4.9059 3.9333 9.1936 3.5649 8.2941

9355 .1823

.7125 10.5895

4847 0569

5.6693 14.6586

.0310 9000 .0002 .9118 .0271 <.0001 <.0001

-4.203

<.0001 <.0001 <.0001 <.0001

109.9939

4.4422 5.3922 - .2022

.1516

14.9873 3.1408 41.8644

2331 5124

12.1061 4.858

7009 .0239 .6266 2950

3.7395

.0463 3445 .0065

35.6368 7.0193

10.6788

.1413

34.9414 87.5113

01439920 01438700 01439570

Little Flat Br

Raymondskill Cr

100.7081

98.2361 07.6758

> 8 18 18 19

Hornbecks Creek

Little Bush Kill

Dingmans Creek

001 001 00 00 00 00

8 8 18 18 19

01439500 01438892 01439092 01439680

Bush Kill

8.585

<.0001 <.0001

Equations relating dissolved oxygen concentration, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., July-September, 2002-04. Table 4-1.

[Relation takes the following form:

$$C = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172*jday)) + (B_3 * cos(0.0172*jday)) \right]$$

C = dissolved oxygen concentration, in percent of saturation;

Q = stream discharge, in cubic feet per second;

 $B_{i} = \text{coefficient for logarithm of stream discharge};$

jday = Julian day of the year; sin = sine; and cos = cosine. B_3 = second seasonal coefficient; B_2 = first seasonal coefficient; log = base-10 logarithm;

č	Station	2	Detected values regression	ected values in regression	-	B	B	~ _	_	\mathbf{B}_2		B	
	number	Z	Number	Number Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	ocale
Adams Creek	01438754	18	18	100	97.9116	<0.0001	4.3532	0.0111	8.5492	0.3295	3.231	0.7209	4.2394
Big Flat Brook	01439830	18	18	100	102.5735	<.0001	5464	.8229	6.8801	.4853	3.5665	.7300	5.1012

8.2941	3.4981	3.7986	5.5081	3.8837	11.5283
.2950	.0369	.6429	.8431	.5303	.8925
21.4644	15.6346	-3.6683	2.2491	-4.7782	3.2063
.0065	.0102	.9266	.6443	.6031	.3663
46.728	19.1288	6953	4.7835	-3.824	21.7004
<.0001	.8527	.9539	.3206	.0146	<.0001
24.2388	2982	.1562	3.0567	3.9386	29.7678
<.0001	<.0001	<.0001	<.0001	<.0001	.0849
123.5643	119.1743	93.7455	96.732	85.3022	55.7299
100	100	100	100	100	100
14	18	18	19	18	18
14	18	18	19	18	18
01439570	0143839602	01438400	01439400	01440100	01438301
Sand Hill Creek	Sawkill Creek	Shimers Brook	Toms Creek	Vancampens Br	Vandermark Cr ¹ 01438301

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek

Table 4-2. Equations relating specific conductance, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

$$log(C) = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172*jday)) + (B_3 * cos(0.0172*jday)) \right]$$

where

Q = stream discharge, in cubic feet per second; C = specific conductance, in microsiemens per centimeter at 25 degrees Celsius;

 B_o = intercept;

 $B_{j} = \text{coefficient for logarithm of stream discharge;}$

log = base-10 logarithm;

 B_{3} = second seasonal coefficient; B_2 = first seasonal coefficient;

jday =Julian day of the year; sin = sine; and

cos = cosine.

N, number of measurements included in regression; Outliers, the number of measurements not included in regression; Percent, percent of number of values included in regression; Significance, level of significance indicating intercept or coefficient is different from zero; <, less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation]

6	Station	Z	Detected in regres	values ession	7	8	8		8	81	7	50°	-
	number	(Outliers)	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	ocale
Adams Creek	01438754	43	43	100	1.872	<0.0001	-0.025	0.0235	0.0328	0.0011	0.0133	0.2579	0.0345
Big Flat Brook	01439830	43	43	100	2.2435	<.0001	-0.1806	<.0001	.0288	.0932	.0255	.1652	.0555
Bush Kill	01439500	43	43	100	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Dingmans Creek	01438892	43	43	100	2.0423	<.0001	-0.1169	<.0001	.0463	<.0001	.0036	.7511	.0357
Hornbecks Creek	01439092	43	43	100	2.2069	<.0001	-0.1652	<.0001	.0327	.0190	0441	.0062	.0482
Little Bush Kill	01439680	43(1)	43	100	1.8125	<.0001	-0.1045	<.0001	6900.	.6631	0569	.0014	.0564
Little Flat Br	01439920	4	44	100	2.677	<.0001	-0.1408	<.0001	6000	7656.	0065	.7155	.0603
Raymondskill Cr	01438700	44	44	100	2.055	<.0001	-0.0388		.03	.0056	0114	.3636	.0393
Sand Hill Creek	01439570	37	37	100	2.4698	<.0001	-0.1049	<.0001	0021	.8878	0257	.1481	.0512
Sawkill Creek	0143839602	44	44	100	2.323	<.0001	-0.1686	<.0001	.0065	9095.	0413	6000	.0397
Shimers Brook	01438400	44	44	100	2.7194	<.0001	-0.3064	<.0001	0299	.0570	0123	.4369	.0523
Toms Creek	01439400	44	44	100	1.9664	<.0001	-0.0501		.0083	.5221	0376	6200.	.0429
Vancampens Br	01440100	40	40	100	1.999	<.0001	-0.1921		.0085	.5943	0037	9208.	.0494
Vandermark Cr1	01438301	44	4	100	2.3431	<.0001	-0.2092	<.0001	.0129	.5072	.004	.8435	.0631

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek. Ý

Table 4-3. Equations relating pH, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

[Relation takes the following form:

$$C = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172*jday)) + (B_3 * cos(0.0172*jday)) \right]$$

C = pH, in standard units;

 $B_0 = \text{intercept};$ $B_1 = \text{coefficient}$ for logarithm of stream discharge;

Q = stream discharge, in cubic feet per second; jday = Julian day of the year;

sin = sine; and cos = cosine. B_3 = second seasonal coefficient; B_2 = first seasonal coefficient; log = base-10 logarithm;

	V		
	zero		
	rom		
٠	Ξ		
	en		
ر	itteren		
	S		
	_		
	ficient		
;	\overline{z}		
ز	coeffic		
	<u>ა</u>		
	intercept or c		
	cept	•	
	Sic		
٠	Ĭ		
	얼)	
•	ndicatin		
:	gic		
	ce I		
	ä		
:	Significan		
	gua)	
٠	_		
	5		
	evel		
-	<u>e</u>		
	tion; Significance,		
	can		
:	₫		
	티)	
ζ	5		
	on;		
•	SSI		
	egressi	`	
	ie)	
	Ξ		
•	ed in		
•	ded		
•	Inded		
•	Inded		
-	ies included		
-	ies included		
	of values included	[uo	
	of values included	ation]	
	per of values included	qua	
	of values included	f equa	
	of number of values included	of equa	
	of number of values included	of equa	
	sent of number of values included	of fit" of equa	
	cent of number of values included	of equa	
	t, percent of number of values included	dness of fit" of equa	
	ent, percent of number of values included	oodness of fit" of equa	
	t, percent of number of values included	dness of fit" of equa	
	Percent, percent of number of values included	oodness of fit" of equa	
	Percent, percent of number of values included	oodness of fit" of equa	
	Percent, percent of number of values included	oodness of fit" of equa	
	ent, percent of number of values included	sure of "goodness of fit" of equa	
	regression; Percent, percent of number of values included	sure of "goodness of fit" of equa	
	d in regression; Percent, percent of number of values included	measure of "goodness of fit" of equa	
	regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	cluded in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	s included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	s included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	s included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	s included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	s included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	s included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	er of measurements included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	ber of measurements included in regression; Percent, percent of number of values included	is a measure of "goodness of fit" of equa	
	er of measurements included in regression; Percent, percent of number of values included	than; n.d., not determined; Scale is a measure of "goodness of fit" of equa	
	umber of measurements included in regression; Percent, percent of number of values included	than; n.d., not determined; Scale is a measure of "goodness of fit" of equa	
	umber of measurements included in regression; Percent, percent of number of values included	than; n.d., not determined; Scale is a measure of "goodness of fit" of equa	

Short name	Station	Z	Detected va regressi	values in ssion	F	60	9	~~ [~]	89	. ~	89	a. ⁽⁰⁾	o S
	number	:	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	
Adams Creek	01438754	43	43	100	6.8398	<0.0001	-0.3824	0.0001	0.2329	0.0103	-0.0027	0.9779	0.3085
Big Flat Brook	01439830	4	44	100	8.0396	<.0001	4695	<.0001	.1807	.0399	.0255	.7752	.2881
Bush Kill	01439500	4	44	100	7.7725	<.0001	5978	.0001	0023	8586.	2775	.0466	.4464
Dingmans Creek	01438892	4	44	100	6866'9		3114	.0003	.0206	.8175	2594	.0046	.3052
Hornbecks Creek	01439092	4	44	100	6.5365	<.0001	1946	.0278	.0377	.7222	8090	.6011	.3717
Little Bush Kill	01439680	4	44	100	7.5667	<.0001	4717	<.0001	.3121	.0002	.062	.5044	.2945
Little Flat Br	01439920	4	44	100	n.d.		n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Raymondskill Cr	01438700	45	45	100	6.8441	<.0001	2169	.0139	.1037	.2022	1696	0589	.2987
Sand Hill Creek	01439570	40	40	100	7.6376		.2886	8000°	.1034	.2089	1072	.2409	.286
Sawkill Creek	0143839602	4	44	100	7.6939	<.0001	455	.0013	.0391	.7268	07	.5779	.4011
Shimers Brook	01438400	43	43	100	8.2925	<.0001	302	.0015	.0141	.7968	1213	.0311	.1822
Toms Creek	01439400	45	45	100	7.1598	<.0001	232	.0871	.0516	.6279	1143	.3061	.3548
Vancampens Br	01440100	40	40	100	7.4473	<.0001	4888	<.0001	.1077	.3270	1208	.2447	.3393
Vandermark Cr	01438301	43	43	100	7.4946	<.0001	3101	.1123	.1229	.3666	1394	.3099	.4292

Table 4-4. Equations relating attenuation turbidity, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

$$\log(C) = B_0 + \left[B_1 * \log(Q) \right] + \left[(B_2 * \sin(0.0172*jday)) + (B_3 * \cos(0.0172*jday)) \right]$$

where

C= attenuation turbidity, in attenuation units; Q= stream discharge, in cubic feet per second; $B_o=$ intercept; Q= stream discharge; Q= stream dischar

 B_2 = first seasonal coefficient; B_3 = second seasonal coefficient;

sin = sine; and cos = cosine.

log = base-10 logarithm;

N, number of measurements included in regression; Outliers, the number of measurements not included in regression; Percent, percent of number of values included in regression; Significance, level of significance indicating intercept or coefficient is different from zero; <, less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation]

ā	Station	z	Detected regre	Detected values in regression	ш	••°		8 -		B ²		a ^s	-
	number	(Outliers)	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Scale
Adams Creek	01438754	43	43	100	0.7318	<0.0001	0.1265	<0.0001	-0.0478	0.0871	-0.1419	<0.0001	0.0912
Big Flat Brook	01439830	42	42	100	.3064	.0003	3805	<.0001	1583	.0001	1891	<.0001	.1267
Bush Kill	01439500	43	43	100	3086	<.0001	.3225	<.0001	1391	<.0001	14	<.0001	.0491
Dingmans Creek	01438892	42(1)	42	100	.6507	<.0001	.2808	<.0001	1081	<.0001	0895	.0005	.0845
Hornbecks Creek	01439092	43	43	100	.8912	<.0001	.0778	.0039	0644	.0579	8860	.0058	.1125
Little Bush Kill	01439680	43	43	100	.6012	<.0001	.3813	<.0001	1329	<.0001	0939	.0102	.114
Little Flat Br	01439920	41	41	100	.6747	<.0001	.281	<.0001	1157	.0020	1439	<.0001	.1158
Raymondskill Cr	01438700	42	42	100	.7238	<.0001	.2771	<.0001	0768	.0239	0815	.0244	.1165
Sand Hill Creek	01439570	38	38	100	.9038	<.0001	.216	<.0001	0143	.6816	1049	.0070	.1162
Sawkill Creek	0143839602	43	43	100	.269	.0019	.3737	<.0001	159	<.0001	1085	.0124	.1363
Shimers Brook	01438400	42	42	100	.6169	<.0001	.4491	<.0001	058	.1163	0984	8900.	.1167
Toms Creek	01439400	43	43	100	.466	<.0001	.2258	<.0001	0558	.2006	0532	.2186	.1341
Vancampens Br	01440100	38	38	100	.6234	<.0001	.1541	<.0001	.0158	.5711	0501	.0460	80.
Vandermark Cr1	01438301	42	42	100	.126	.5085	.3782	<.0001	0589	.3536	074	.2466	.1947

^{&#}x27;Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

Equations relating total phosphorus, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., Table 4-5.

$$log(C) = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172*jday)) + (B_3 * cos(0.0172*jday)) \right]$$

C = total phosphorus, in milligrams per liter as P; $B_o = \text{intercept};$

 $B_{i} = \text{coefficient for logarithm of stream discharge};$

 B_2 = first seasonal coefficient;

log = base-10 logarithm;

 \vec{B}_3 = second seasonal coefficient;

Q = stream discharge, in cubic feet per second; jday = Julian day of the year;

sin = sine; and cos = cosine.

N, number of measurements included in regression; Percent, percent of number of values included in regression; Significance, level of significance indicating intercept or coefficient is different from zero; <, less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation]

Short name	Station	z	Detected valu regressior	values in ssion	ш	e - 0	ш	~ -	m	٥	••	~ "	Scale
	number	<u> </u>	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	
Adams Creek	01438754	44	44	100	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Big Flat Brook	01439830	44	44	100	-2.5998	<0.0001	.365	<0.0001	-0.2551	<0.0001	- 0.331	<0.0001	0.1806
Bush Kill	01439500	4	44	100	5587	.0064	4188	<.0001	029	.6682	0458	.5329	.2352
Dingmans Creek	01438892	4	44	100	-2.0864	<.0001	.1406	<.0001	1257	<.0001	2504	<.0001	.0901
Hornbecks Creek	01439092	4	44	100	-1.9772	<.0001	.057	.1911	1453	.0055	3213	<.0001	.1832
Little Bush Kill	01439680	4	44	100	-2.0826	<.0001	.165	<.0001	1406	<.0001	2633	<.0001	.0937
Little Flat Br	01439920	4	44	100	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Raymondskill Cr	01438700	45	45	100	-1.8408	<.0001	0371	.2211	1436	<.0001	2244	<.0001	.1026
Sand Hill Creek	01439570	40	40	100	-1.7173	<.0001	.0407	.4200	1106	.0213	3916	<.0001	.1669
Sawkill Creek	0143839602	4	44	100	-2.2994	<.0001	.2574	<.0001	2279	<.0001	2499	<.0001	.1624
Shimers Brook	01438400	4	44	100	-2.0141	<.0001	.2019	6900°	1066	.0157	2145	<.0001	.1467
Toms Creek	01439400	45	45	100	-2.0883	<.0001	2900.	.9279	1575	<i>L</i> 900.	1648	8900.	.1936
Vancampens Br	01440100	39	39	100	-2.2606	<.0001	.0787	.1056	1455	.0015	1827	<.0001	.1413
Vandermark Cr1	01438301	44	44	100	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek

Table 4-6. Equations relating dissolved orthophosphate phosphorus, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

$$log(C) = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172*jday)) + (B_3 * cos(0.0172*jday)) \right]$$

where

C = dissolved orthophosphate phosphorus, in milligrams per liter as P;

 B_0 = intercept;

 B_{1}^{o} = coefficient for logarithm of stream discharge;

Q = stream discharge, in cubic feet per second; $jd\alpha y$ = Julian day of the year; sin = sine; and

ssion: Significance level of significance indicating int cos = cosine. mber of values included in B_2' = first seasonal coefficient; B_3 = second seasonal coefficient; log = base-10 logarithm;

ć,	
ver of significance maleaning intercept of coefficient is unferent from zero,	
it of number of values included in regression, significance, reve of fit" of equation]	
c, humoel of measurements included in regression, reacent, percent < less than; n.d., not determined; Scale is a measure of "goodness of "goodness" of	

Ohor to do	Station	2	Detected values in regression	l values ession	ш	••	_ -	B		32	a	.es	0
	number	Z	Number Percer	Percent	Value	Signifi- cance	Value	Signifi- Value cance	Value	Signifi- cance	Value	Signifi- cance	ocale
Adams Creek	01438754	43	10	23	n.d.	n.d.		n.d.	n.d.		n.d.	n.d.	n.d.
Big Flat Brook	01439830	43	3	7	n.d.	n.d.		n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Bush Kill	01439500	4	33	75	0.5188	0.0605		<0.0001	0.0268	0.7584	0.1892	0.0509	0.2776
Dingmans Creek	01438892	4	14	32	-2.162	<.0001		.0007	600.	.9130	2354	.0062	.1461
Hornbecks Creek	01439092	4	4	6	n.d.	n.d.		n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Little Bush Kill	01439680	4	22	50	-2.1484	<.0001		.0034	0926	.0201	2626	<.0001	.1029
Little Flat Br	01439920	4	26	59	-2.9465	<.0001		.0007	3175	.0002	4034	<.0001	.2359
Raymondskill Cr	01438700	45	11	24	n.d.	n.d. n.d.		n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Sand Hill Creek	01439570	39	30	77	-2.2787	<.0001		.8405	1585	0900	4998	<.0001	.1801
Sawkill Creek	0143839602	4	28	49	-2.4584	<.0001		.3321	2715	<.0001	2856	<.0001	.1364
Shimers Brook	01438400	4	16	36	-2.4748	<.0001		.3149	2507	.0260	2697	.0178	.2482
Toms Creek	01439400	45	23	51	-2.2363	<.0001		<.0001	1553	.0002	2157	<.0001	.0934
Vancampens Br	01440100	39	10	26	n.d.	n.d.		n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Vandermark Cr ¹	01438301	44	41	93	-2.0571		.0173	.8155	1604	.0017	2163 <.0001	<.0001	.161

^{&#}x27;Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

Equations relating total nitrogen, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., Table 4-7.

$$log(C) = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172*jday)) + (B_3 * cos(0.0172*jday)) \right]$$

C = total nitrogen, in milligrams per liter as N;

 $B_o = \text{intercept};$

 $B_i = \text{coefficient for logarithm of stream discharge};$

Q = stream discharge, in cubic feet per second; jday =Julian day of the year; sin = sine; and

N, number of measurements included in regression; Outliers, the number of measurements not included in regression; Percent, percent of number of values included in regression; Significance, level of significance indicating intercept or coefficient is different from zero; <, less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation] B_2' = first seasonal coefficient; B_3' = second seasonal coefficient; log = base-10 logarithm;

400	Station	z	Detected value in regression	values	<u></u>	~		8-	m	2	<u></u>	a [°]	200
	number	(Outliers)	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	ocale
Adams Creek	01438754	4	44	100	-0.743	<0.0001	-0.0564	0.0875	-0.0816	090000	-0.2607	<0.0001	0.103
Big Flat Brook	01439830	4	42	95	-1.1805	<.0001	.25	<.0001	1785	<.0001	2615	<.0001	.1433
Bush Kill	01439500	4	44	100	142	.3514	2027	6000.	0912	6690.	1026	.0604	.1749
Dingmans Creek	01438892	44	44	100	6352	<.0001	.0575	.0052	0429	.0459	1357	<.0001	.0737
Hornbecks Creek	01439092	43(1)	43	100	6558	<.0001	.0722	.0013	0532	.0491	1561	<.0001	.094
Little Bush Kill	01439680	44	44	100	7749	<.0001	.1401	<.0001	1242	<.0001	1691	<.0001	6680.
Little Flat Br	01439920	44	44	100	4339		.1083	6000	0873	.0002	1038	<.0001	.077
Raymondskill Cr	01438700	45	45	100	7352	<.0001	880.	<.0001	1061	<.0001	1375	<.0001	.0727
Sand Hill Creek	01439570	40	40	100	3509		.028	.2982	05	.0515	1459	<.0001	.0891
Sawkill Creek	0143839602	44	44	100	3496	<.0001	0765	.0370	1756	<.0001	1323	<.0001	.1042
Shimers Brook	01438400	44	44	100	5178	<.0001	.2126	<.0001	0447	.0378	0425	.0501	.0715
Toms Creek	01439400	45	45	100	7673	<.0001	.0339	.5327	0581	.1739	1276	.0044	.1423
Vancampens Br	01440100	38(1)	27	72	-1.1177	<.0001	.0138	.8183	8000	.9891	2123	<.0001	.1611
Vandermark Cr ¹	01438301	43(1)	43	100	.0314	.7236	2058	<.0001	0719	.0092	0851	.0030	6680.

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

Table 4-8. Equations relating dissolved nitrate plus nitrite, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

$$\log(C) = B_0 + \left\lceil B_1 * \log(Q) \right\rceil + \left\lceil (B_2 * \sin(0.0172*jday)) + (B_3 * \cos(0.0172*jday)) \right\rceil$$

where

C= dissolved nitrate plus nitrite, in milligrams per liter as N;

B0 = intercept;

BI =coefficient for logarithm of stream discharge;

B2 = first seasonal coefficient;

B3 = second seasonal coefficient;

sin = sine; and cos = cosine.

Q = stream discharge, in cubic feet per second;

jday =Julian day of the year;

N, number of measurements included in regression; Outliers, the number of measurements not included in regression; Percent, percent of number of values included in regression; Significance, level of log = base-10 logarithm;

significance indicating intercept or coefficient is different from zero; <, less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation]

1	Station	2	Detected values in regression	values in ssion	_	മ°	-	æ ⁻	ш	2 2	_	മ്	9
	number	(Outliers)	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	ocale
Adams Creek	01438754	43	32	74	-1.1406	<0.0001	-0.3591	<0.0001	0.0306	0.6147	-0.3077	<0.0001	0.1955
Big Flat Brook	01439830	43(1)	27	49	-1.0354	<.0001	3071	.0064	0137	.8820	2049	.0242	.2677
Bush Kill	01439500	4	27	61	1.3704	.0002	-1.141	<.0001	.2244	.0624	7777	.0328	.3476
Dingmans Creek	01438892	4	42	95	7641	<.0001	2919	<.0001	1960.	.0750	1235	.0263	.185
Hornbecks Creek	01439092	43(1)	40	93	-1.149	<.0001	.0383		.1335	.0483	085	.2491	.2349
Little Bush Kill	01439680	4	37	84	8238	<.0001	2758		.0016	7086.	1842	.0104	.2252
Little Flat Br	01439920	44	44	100	5321	<.0001	0499		0503	.3053	0246	.6024	.1608
Raymondskill Cr	01438700	45	35	78	-1.3306	<.0001	0239		0141	.7808	.0455	.4136	.1823
Sand Hill Creek	01439570	39	39	100	7082	<.0001	1154	.0617	0441	.4538	0709	.2770	.2038
Sawkill Creek	0143839602	44	44	100	1043	.1028	3877	<.0001	1402	<.0001	0823	.0103	.1023
Shimers Brook	01438400	44	44	100	6511	<.0001	9/90	.3538	.0049	8806.	.0841	.0527	.1431
Toms Creek	01439400	45	45	100	7905	<.0001	1548	.0138	9200.	.8771	1605	.0019	.1645
Vancampens Br	01440100	39	16	41	-1.2883	<.0001	3986	.0042	.1381	.2907	1915	.1426	.3028
Vandermark Cr1	01438301	43(1)	43	100	.1373	.1519	3192	<.0001	0597	.0451	0597	.0530	6960:

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

Table 4-9. Equations relating dissolved ammonia, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and

$$\log(C) = B_0 + \left[B_1 * \log(Q) \right] + \left[(B_2 * \sin(0.0172*jday)) + (B_3 * \cos(0.0172*jday)) \right]$$

C = dissolved ammonia, in milligrams per liter as N;

 $B_o = \text{intercept};$

Q = stream discharge, in cubic feet per second; jday = Julian day of the year; sin = sine; and cos = cosine. $B_I = \text{coefficient}$ for logarithm of stream discharge; $B_2 = \text{first seasonal coefficient}$; $B_3 = \text{second seasonal coefficient}$; Iog = base-10 logarithm;

less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation]	Detected values in B ₁ B ₂ B ₃ Station Legression B ₀ B ₁ B ₃	number N Scale
less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation]		Short lidille

100	Station	2	Detected values in regression	values in ssion		ຕິ	_	~ -		2	Δ.	ຸຕ	6
	number	2	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	ocale
Adams Creek	01438754	43	14	33	-2.5232	<0.0001	-0.0581	0.5948	-0.0684	0.5022	-0.4705	0.0088	0.2522
Big Flat Brook	01439830	4	4	6	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Bush Kill	01439500	4	13	30	77	.1811	8053	.0013	.0048	6626	0569	.7908	.4484
Dingmans Creek	01438892	4	14	32	-2.4	<.0001	.0238	<i>6LLL</i> :	.0357	.6475	1271	.1416	.2089
Hornbecks Creek	01439092	4	27	61	-1.9595	<.0001	2654	.0001	.144	.0928	1869	.0343	.2542
Little Bush Kill	01439680	4	16	36	-2.7343	<.0001	.1441	.1724	2432	.0212	2425	.0339	.2782
Little Flat Br	01439920	4	33	75	-1.8842	<.0001	1985	.0246	.1354	.0349	188	.0029	.196
Raymondskill Cr	01438700	45	27	09	-1.8688	<.0001	2258	.0005	0194	.7452	1446	.0297	.1918
Sand Hill Creek	01439570	39	23	59	-2.2382	<.0001	.0615	.4642	.0581	.4726	1994	.0318	.2605
Sawkill Creek	0143839602	4	7	16	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Shimers Brook	01438400	4	15	34	-2.7444	<.0001	.5269	.0023	7600	9568.	.1094	.1395	.1974
Toms Creek	01439400	45	2	4	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Vancampens Br	01440100	38	1	3	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Vandermark Cr ¹	01438301	44	5	11	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

Table 4-10. Equations relating acid-neutralizing capacity, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

$$log(C) = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172*jday)) + (B_3 * cos(0.0172*jday)) \right]$$

Q = stream discharge, in cubic feet per second; jday = Julian day of the year; sin = sine; and cos = cosine. C = acid-neutralizing capacity, in milligrams per liter as CaCO₃; $\vec{B_l} = \text{coefficient for logarithm of stream discharge}$; $\vec{B_3}$ = second seasonal coefficient; B_{i} = first seasonal coefficient;

log = base-10 logarithm;

N, number of measurements included in regression; Percent, percent of number of values included in regression; Significance, level of significance indicating intercept or coefficient is different from zero; <, less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation]

Short name	Station	2	Detected values in regression	values in ssion		e e		8 -	u	3	_	a [°]	oleo
	number	2	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	
Adams Creek	01438754	18	18	100	0.9836	<0.0001	-0.1839		-0.0618	0.0010	-0.0791	<0.0001	0.0437
Big Flat Brook	01439830	18	18	100	1.6657	<.0001	3036	<.0001	0416	9000	0283	.0361	.0283
Bush Kill	01439500	19	19	100	1.3111		2172	<.0001	0167	.4636	0677	6800.	.055
Dingmans Creek	01438892	18	18	100	1.1929	<.0001	2002		77770	<.0001	0957	<.0001	.0419
Hornbecks Creek	01439092	18	18	100	1.0421	<.0001	0549	.0202	9880	.0004	0658	.0138	.0593
Little Bush Kill	01439680	18	18	100	8666		1474	<.0001	6090	.0033	1066	<.0001	.0524
Little Flat Br	01439920	18	18	100	2.2545		2103	<.0001	0043	.8125	.0133	.4940	.043
Raymondskill Cr	01438700	18	18	100	1.0979	<.0001	1388	.0003	1159	<.0001	1374	<.0001	.0742
Sand Hill Creek	01439570	16	16	100	2.0058		0391	.1383	0144	.5814	0465	.0948	.0518
Sawkill Creek	0143839602	19	19	100	1.2588	<.0001	1405		0885	<.0001	0601	<.0001	.0315
Shimers Brook	01438400	18	18	100	2.2961	<.0001	3453	<.0001	0506	.0220	0305	.2088	.0529
Toms Creek	01439400	19	19	100	1.2215	<.0001	1595	<.0001	0501	<.0001	0519	<.0001	.027
Vancampens Br	01440100	15	15	100	1.5782	<.0001	334	<.0001	0113	.6514	0168	.4716	.0458
Vandermark Cr ¹	01438301	18	18	100	1.2575	<.0001	1586	<.0001	072	<.0001	0527	.0004	.0307

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

Table 4-11. Equations relating dissolved chloride, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

$$log(C) = B_0 + \left[B_1 * log(Q) \right] + \left[(B_2 * sin(0.0172*jday)) + (B_3 * cos(0.0172*jday)) \right]$$

where

C = dissolved chloride, in milligrams per liter as Cl;

Q = stream discharge, in cubic feet per second;

 $B_o = \text{intercept};$

 $\vec{B_j}$ = coefficient for logarithm of stream discharge;

N, number of measurements included in regression; Percent, percent of number of values included in regression; Significance, level of significance indicating intercept or coefficient is different from zero; <, *jday* = Julian day of the year; sin = sine; and B_3 = second seasonal coefficient; B_2' = first seasonal coefficient; log = base-10 logarithm;

less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation]

č	Station	:	Detected regres	d values in ression	ш	B	m	-	B	2	B	m	-
Short name	number	Z	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Scale
Adams Creek	01438754	18	18	100	1.0199	<0.0001	0.0129	0.6939	0.052	0.0481	0.0012	0.9661	0.0612
Big Flat Brook	01439830	18	18	100	1.2582	<.0001	1461	.0109	9690.	.1374	.0761	.1456	.1097
Bush Kill	01439500	19	19	100	1.1771	<.0001	2067	<.0001	.0711	.0043	.0329	.2452	.0602
Dingmans Creek	01438892	18	18	100	1.236	<.0001	1341	<.0001	.0805	<.0001	0015	.9408	.0447
Hombecks Creek	01439092	18	18	100	1.5071	<.0001	2209	<.0001	.0433	.0873	6690	.0093	9650.
Little Bush Kill	01439680	18	18	100	.9137	<.0001	1171	<.0001	.0636	.0003	0169	.4116	.044
Little Flat Br	01439920	18	18	100	1.7199	<.0001	1421	.0007	.0235	.4227	.0377	.2310	.07
Raymondskill Cr	01438700	18	18	100	1.3065	<.0001	0179	.4543	.0619	8000	026	.1983	.0462
Sand Hill Creek	01439570	16	16	100	1.4178	<.0001	2662	<.0001	.1186	9900.	0384	.4103	.0867
Sawkill Creek	0143839602	19	19	100	1.6453	<.0001	2053	<.0001	.0559	.0002	0392	.0273	.038
Shimers Brook	01438400	18	18	100	1.635	<.0001	2895	<.0001	0546	.0912	0145	.6841	.0774
Toms Creek	01439400	19	19	100	1.073	<.0001	0548	.2461	.078	.0204	0354	.3424	.0784
Vancampens Br	01440100	15	15	100	5907	<.0001	151	<.0001	6920.	.0015	0019	.9311	.0442
Vandermark Cr1	01438301	18	18	100	1.5601	<.0001	2271	.0187	.0385	.5312	.0984	.1093	.1277

'Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek

Table 4-12. Equations relating dissolved calcium, streamflow, and season, at stations on streams in and near the Delaware Water Gap National Recreation Area, Pa. and N.J., 2002-04.

$$\log(C) = B_0 + \left[B_1 * \log(Q) \right] + \left[(B_2 * \sin(0.0172*jday)) + (B_3 * \cos(0.0172*jday)) \right]$$

C = dissolved calcium, in milligrams per liter as Ca; $B_o = \text{intercept};$

 $B_{i} = \text{coefficient for logarithm of stream discharge};$

 B_2' = first seasonal coefficient; B_3' = second seasonal coefficient;

jday = Julian day of the year; sin = sine; and

Q = stream discharge, in cubic feet per second;

cos = cosine.

N, number of measurements included in regression; Percent, percent of number of values included in regression; Significance, level of significance indicating intercept or coefficient is different from zero; <, less than; n.d., not determined; Scale is a measure of "goodness of fit" of equation] log = base-10 logarithm;

ā	Station	:	Detected values in regression	values in ssion	•	മ°	a	-	m	B ²	മ്	8	-
У пот пате	number	Z	Number	Percent	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Value	Signifi- cance	Scale
Adams Creek	01438754	18	18	100	0.6401	<0.0001	-0.0842	<0.0001	-0.0122	0.2181	-0.0325	0.0019	0.023
Big Flat Brook	01439830	18	18	100	1.1667	<.0001	2195	<.0001	0082	9029.	.0102	.6380	.0453
Bush Kill	01439500	19	19	100	.843	<.0001	1342	<.0001	0121	.4469	0324	.0749	.0386
Dingmans Creek	01438892	18	18	100	.7748	<.0001	1203	<.0001	0162	.2271	0385	.0033	.0296
Hornbecks Creek	01439092	18	18	100	.8463	<.0001	0959	<.0001	0178	.2808	0641	.0003	.0388
Little Bush Kill	01439680	18	18	100	.6047	<.0001	0642	<.0001	0294	.0243	0567	.0002	.0329
Little Flat Br	01439920	18	18	100	1.7704	<.0001	1716	<.0001	0074	.7211	.0329	.1411	.0496
Raymondskill Cr	01438700	18	18	100	.7331	<.0001	0547	.0259	0418	.0274	0644	.0018	.0473
Sand Hill Creek	01439570	16	16	100	1.7004	<.0001	1016	<.0001	.0139	.3342	0013	.9311	.0286
Sawkill Creek	0143839602	19	19	100	1.0985	<.0001	1889	<.0001	0125	.2378	0293	.0208	.0271
Shimers Brook	01438400	18	18	100	1.8072	<.0001	3039	<.0001	0486	.0210	0208	.3683	.0504
Toms Creek	01439400	19	19	100	.8432	<.0001	0855	<.0001	0003	.9774	0251	.0453	.0263
Vancampens Br	01440100	15	15	100	1.0503	<.0001	2563	<.0001	0033	.8564	0055	.7460	.0333
Vandermark Cr ¹	01438301	18	18	100	1.0425	<.0001	1745	<.0001	023	.2395	.0045	.8200	.0407

^{&#}x27;Streamflow for relations for Vandermark Creek (station 01438301) is daily mean streamflow measured at Flat Brook near Flatbrookville (station 01440000) on the days the samples were collected from Vandermark Creek.

For additional information, write to: Director U.S. Geological Survey New Jersey Water Science Center Mountain View Office Park 810 Bear Tavern Rd., Suite 206 West Trenton, NJ 08628

or visit our Web site at: http://nj.usgs.gov/